www.rsc.org/obc

Asymmetric synthesis of the *cis*- and *trans*-stereoisomers of 4-aminopyrrolidine-3-carboxylic acid and 4-aminotetrahydrofuran-3carboxylic acid

Mark E. Bunnage,^{*a*} Stephen G. Davies,^{**b*} Paul M. Roberts,^{*b*} Andrew D. Smith^{*b*} and Jonathan M. Withey^{*b*}

^a Discovery Chemistry, IPC 675, Pfizer Global Research and Development, Sandwich, Kent, UK CT13 9NJ

^b Department of Organic Chemistry, University of Oxford, Chemistry Research Laboratory, Mansfield Road, Oxford, UK OX1 3TA

Received 19th May 2004, Accepted 28th July 2004 First published as an Advance Article on the web 1st September 2004

The diastereoselective conjugate addition of lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide has been successfully applied to the first asymmetric syntheses of *cis*-(3*S*,4*R*)- and *trans*-(3*R*,4*R*)-4-aminotetrahydrofuran-3-carboxylic acids (26% and 25% overall yield respectively, >98% d.e. and >97% e.e. in each case). Furthermore, the most efficient asymmetric synthesis to date of *cis*-(3*R*,4*R*)- and *trans*-(3*R*,4*S*)-4-aminopyrrolidine carboxylic acids is delineated: for *cis*-(3*R*,4*R*), four steps, >98% d.e., 52% overall yield; for *trans*-(3*R*,4*S*), five steps, >98% d.e., 50% overall yield.

Introduction

Heterocyclic β -amino acid derivatives of *cis*-(1*R*,2*S*)-2-aminocyclopentane-1-carboxylate 1 (cispentacin) are a medicinally interesting and much studied molecular class. While cispentacin is a potent antifungal agent,¹ the incorporation of an oxygen or sulfur heteroatom within the carbocyclic skeleton to generate the corresponding (\pm) -4-aminotetrahydrofuranand (\pm) -4-aminotetrahydrothiophene-3-carboxylates 2 and 3 results in a significant reduction of biological activity.² However, racemic cis-4-aminopyrrolidine-3-carboxylic acid 4 has been used to probe the structure of the GABA receptor,³ while cis-N-Boc-4-aminopyrrolidine-3-carboxylic acid 5 is a modestly active influenza neuramidase inhibitor. Using N-Boc-5 as a lead structure, Wang et al. have employed a combination of combinatorial chemistry and structure based drug design to discover the trisubstituted pyrrolidine carboxylic acid A-192558 6 as a potent influenza neuramidase inhibitor.⁴ Furthermore, water soluble short chain peptides such as 8 containing both transpentacin and trans-(3R,4S)-4-aminopyrrolidine-3carboxylic acid 7 residues have been shown to adopt distinct helical secondary structures in solution (Fig. 1).5

Despite recent interest in the preparation of this structural class, only limited methods for the asymmetric synthesis of these heterocyclic β-amino acid derivatives in enantiomerically pure form exist. For instance, the asymmetric synthesis of cis-4-aminopyrrolidine-3-carboxylic acid derivatives has been achieved using enantioselective hydrogenation⁶ and 1,3-dipolar cycloaddition of nitrones,^{7,8} while the synthesis of *trans*-4-aminopyrrolidine-3-carboxylic acid has been achieved diastereoselectively using radical addition9 and in an asymmetric fashion through reductive amination.¹⁰ However, there is no single methodology that allows the selective asymmetric synthesis of both cisand trans-diastereoisomers. Previous investigations from this laboratory have shown that conjugate addition of homochiral lithium (S)-N-benzyl-N- α -methylbenzylamide 9 to tert-butyl cyclopentene-1-carboxylate 10 and subsequent N-deprotection and ester hydrolysis allows an efficient synthesis of cispentacin **12**. Selective epimerisation of β -amino ester *cis*-(1*R*,2*S*, α *S*)-11 to the thermodynamic epimer trans- $(1S, 2S, \alpha S)$ -13 leads to (1S,2S)-transpentacin 14 (Scheme 1).¹¹

As an extension of this methodology, it was envisaged that the conjugate addition of homochiral lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide **9** to a range of dihydrofuran, *N*-protected dihydropyrrole and dihydrothiophene α , β -unsaturated esters





would facilitate the asymmetric synthesis of both *cis*- and *trans*stereoisomers of the corresponding β -amino acid derivatives, and we delineate herein our investigations within this area.

Results and discussion

Synthesis of heterocyclic α,β-unsaturated esters

Initial investigations centred upon a simple and general route for the synthesis of a range of *tert*-butyl dihydrofuran, *N*-protected dihydropyrrole and dihydrothiophene α,β -unsaturated esters. As *tert*-butyl (*RS*)-3-alkylcyclopentene-1-carboxylates may be readily prepared from the corresponding β -keto esters,¹² an analogous disconnection in the heterocyclic series leads to the requirement for an efficient synthesis of the corresponding heterocyclic β -keto esters **15**. It was envisaged that this structural framework could be regioselectively assembled in a 'one pot' procedure by a tandem conjugate addition–Dieckmann cyclisation protocol (Fig. 2).



Scheme 1 Reagents and conditions: (i). Lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide 9, THF, -95 °C then 2,6-di-*tert*-butyl phenol; (ii). KO'Bu, 'BuOH, rt; (iii). Pd–C, MeOH, H₂ (5 atm), rt; (iv). TFA then Dowex 50WX8-200.



Fig. 2 Retrosynthetic analysis of *tert*-butyl dihydrofuran, *N*-protected dihydropyrrole and dihydrothiophene α , β -unsaturated esters.

In a model study, and using a modification of the literature procedure,¹³ tert-butyl 4-oxotetrahydrofuran-3-carboxylate 16 was obtained in 68% yield by reaction of the anion of ethyl glycolate with tert-butyl acrylate in DMSO at rt. This experimentally simple procedure was subsequently adapted to the synthesis of tert-butyl 4-oxotetrahydrothiophene-3-carboxylate 22 which was obtained in 81% overall yield as a 46:54 mixture with its enol tautomer 23, and the corresponding pyrrolidine β keto ester which was isolated as its stable potassium enolate salt 19 in 77% yield from the reaction of the anion of N-Boc-glycine methyl ester and *tert*-butyl acrylate.⁴ These heterocyclic β-keto esters 16, 19 and 22:23 were subsequently chemoselectively reduced using either sodium borohydride or sodium cyanoborohydride14 to the corresponding alcohols 17, 20 and 24, which underwent facile dehydration upon treatment with PPh₃/diisopropyl azodicarboxylate (DIAD), furnishing the desired α , β -unsaturated acceptors 18, 21 and 25 (Scheme 2).

Probing the reactivity of heterocyclic α , β -unsaturated acceptors: conjugate addition of lithium dibenzylamide

With efficient routes to heterocyclic α , β -unsaturated esters 18, 21 and 25 in hand, attention turned to studying the conjugate addition reactions of lithium amides to these substrates. In order to quantify the reactivity of the heterocyclic α,β unsaturated esters in this reaction manifold, initial model studies were concerned with the addition of achiral lithium dibenzylamide. Addition of tert-butyl 2,5-dihydrofuran-3-carboxylate 18 to lithium dibenzylamide (1.6 eq) gave a complex mixture of four products 26:27:28:29 in a 17:26:5:52 ratio, with chromatographic purification allowing separation of all four compounds. The expected C(3)-epimeric β -amino esters 26 and 27 were obtained in 10% and 18% isolated yield respectively, and in >98% d.e. in each case. Subsequent treatment of (3SR, 4RS)-26 (>98% d.e.) with KO'Bu in 'BuOH gave quantitatively the thermodynamic β -amino ester (3RS,4RS)-27 (>98% d.e.), allowing assignment of the relative configuration within 26 and 27. The two other products of the conjugate addition reaction were the oligometric β -amino esters 28 and 29, isolated in 4% and 46% yield respectively, which presumably derive from conjugate addition of the *in situ* generated (Z)- β -amino lithium enolate to



Scheme 2 Reagents and conditions: (i). NaH (1.0 eq), Et₂O, rt, 0.5 h then *tert*-butyl acrylate (1.2 eq), DMSO, rt, 0.5 h; (ii). NaBH₄ (1.0 eq), 'PrOH, 0 °C to rt; (iii). PPh₃ (1.5 eq), DIAD (1.3 eq), THF, 0 °C to rt; (iv). *tert*-butyl acrylate (1.0 eq), KO'Bu (1.1 eq), THF, 0 °C, 0.5 h, then rt, 16 h; (v). NaBH₃CN (1.1 eq), MeOH, pH 3–4, rt.

18. The relative configuration at C(3) within oligomeric β-amino esters **28** and **29** was assigned by assuming the known *anti*-facial bias for the reaction of cyclic β-amino enolates derived from lithium amide conjugate addition with electrophiles.¹⁵ The relative *cis*- configuration at C(3') and C(4') within oligomeric β-amino ester **28** was verified independently by ¹H NOE difference analysis, while the epimeric nature of **28** and **29** was confirmed through conversion of (3SR,4RS,3'SR,4'SS)-**28** to the thermodynamic adduct (3SR,4RS,3'SR,4'RS)-**29** (>98% d.e.) upon treatment with KO'Bu in 'BuOH. (Scheme 3).



Scheme 3 Reagents and conditions: (i). lithium dibenzylamide (1.6 eq), THF, -78 °C then NH₄Cl_(aq), -78 °C to rt; (ii). KO'Bu, 'BuOH, Δ , 3 h.

Although low levels of diastereoselectivity upon protonation of the β -amino enolates generated in this reaction with NH₄Cl may be expected,¹¹ the formation of oligomeric products in the reaction was unexpected, as during our extensive research utilising the conjugate addition of lithium amides this reaction manifold has not been identified under these standard conditions. In order to bias the reaction toward the formation of the desired β -amino esters **26** and **27** it was predicted that slow addition of ester **18** to lithium dibenzylamide would disfavour the formation of the oligomeric products. Thus, dropwise addition of a dilute solution of ester **18** to lithium dibenzylamide (1.6 eq) at $-78 \,^{\circ}$ C, followed by addition of NH₄Cl gave a 58:42 mixture of only the C(3)-epimeric β -amino esters **26** and **27**, in 40% and 31% isolated yield respectively. Conversely, addition of ester **18** in THF to 0.5 eq of lithium dibenzylamide gave a 7:93 mixture of only the C(4')-epimeric oligomeric esters **28** and **29** in 7% and 71% isolated yield respectively (Scheme 4).



Scheme 4 Reagents and conditions: (i). lithium dibenzylamide (1.6 eq), **18** (1.0 eq), dilute solution added over 2 h, THF, -78 °C then NH₄Cl_(aq), -78 °C to rt; (ii). lithium dibenzylamide (0.5 eq), **18** (1.0 eq), THF, -78 °C then NH₄Cl_(aq), -78 °C to rt.

Subsequent studies were concerned with evaluating the product distribution arising from addition of tert-butyl 2,5dihydrothiophene-3-carboxylate 25 and tert-butyl N-Boc-2,5dihydropyrrole-3-carboxylate 21 to lithium dibenzylamide. Addition of 25 to lithium dibenzylamide under standard conditions gave a complex mixture of products, which contained the C(3)-epimeric β -amino esters (3RS,4RS)-30 and (3SR,4RS)-31 in a 54:46 ratio, although there was no evidence of the formation of oligomeric products. Chromatographic purification yielded an inseparable mixture of β -amino esters (3RS.4RS)-30 and (3SR.4RS)-31 (54:46) in 49% vield. with treatment of this isolated mixture with KO'Bu in 'BuOH allowing quantitative conversion to (3SR, 4RS)-31 in >98% d.e. Chromatography also allowed the isolation of the deconjugated ester tert-butyl 2,3-dihydrothiophene-3-carboxylate 32 in 25% yield, consistent with γ -deprotonation competing with lithium amide addition in this case. Similarly, addition of 21 to lithium dibenzylamide again gave a complex mixture, containing β -amino ester (3RS,4RS)-33 as the sole C(3)-epimer, with chromatography furnishing rotameric (3RS,4RS)-33 in 41% isolated yield. Attempted conversion of (3RS,4RS)-33 to its C(3)-epimer with KO'Bu in 'BuOH proved unsuccessful, leading instead to extensive decomposition, although an 8.1% NOE enhancement from C(4)-H to C(3)-H was indicative of a syn relationship between the amino and carboxylate functionalities within 33. Chromatographic purification also allowed the isolation and identification of the deconjugated adduct tert-butyl N-Boc-2,3-dihydropyrrole-3-carboxylate 34 in 22% yield and tert-butyl N-Boc pyrrole-1,3-dicarboxylate 35 in 11% yield, again consistent with γ -deprotonation competing with lithium amide addition. It is notable that N-Boc amino protection has been used by Beak et al. as an activating and directing group for lithiation in the deprotonation of a variety of cyclic and allylic amines, which may explain the facile γ -deprotonation observed in this case (Scheme 5).¹⁶

Asymmetric synthesis of heterocyclic β-amino esters: conjugate addition of homochiral lithium *N*-benzyl-*N*-α-methylbenzylamide

With model studies complete, the asymmetric synthesis of the heterocyclic β -amino esters was investigated. Slow addition



of dihydrofuran carboxylate 18 to a solution of homochiral lithium-(S)-N-benzyl-N- α -methylbenzylamide 9 gave a crude reaction mixture containing four products 36:37:38:39 in a 47:4:2:47 ratio. Purification by chromatography on silica gave the separable C(3)-epimeric β -amino esters (3S,4R, α S)-36 and $(3R,4R,\alpha S)$ -37 in 31% and 3% yield respectively. The configuration at C(4) within 36 and 37 relative to the N- α -methylbenzyl stereocentre was assigned by analogy with the model previously used to explain the stereoselectivity observed during addition of lithium amide 9 to α , β -unsaturated esters,¹⁷ with the relative configuration at C(4) and C(3) confirmed by quantitative conversion of 36 to the thermodynamic epimer 37. The oligomeric product 38 was obtained as a minor reaction component in 2% vield,¹⁸ with further elution giving **39**, a product containing two *N*-benzyl-*N*- α -methylbenzylamino functionalities in 14% yield, and the related product 40 in 17% yield. The α , β -unsaturated lactone 40 was not present in the crude reaction mixture, and is presumed to arise from the retro-conjugate addition of Nbenzyl-N-a-methylbenzylamine under the mildly acidic conditions of the purification medium (Scheme 6).¹⁹ The relative configuration within 40 was unambiguously established by single crystal X-ray analysis, with the absolute $(3S, 4R, \alpha S)$ configuration derived from the known configuration of the (S)-N- α -methylbenzyl stereocentre (Fig. 3).

Diamino ester **39** presumably arises as a result of slow ring opening of the initially formed (*Z*)- β -amino enolate **41** arising from conjugate addition of lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide, followed by 5-*exo-trig* ring-closure to give α , β -unsaturated lactone **42**.²⁰ Subsequent diastereoselective conjugate addition of β -amino enolate **41** *anti*- to its stereodirecting amino substituent, and protonation of the resulting enolate *anti*- to the neighbouring C(4)-amino group will furnish the diamino ester **39** (Fig. 4).

On the basis of this mechanistic hypothesis, it was envisaged that minimisation of the undesired oligomeric fragmentation products in this reaction could be achieved provided that the reaction was quenched sufficiently quickly so as to allow conjugate addition to occur, but not promote fragmentation. In this manner, addition of ester **18** to an excess of lithium amide (*S*)-**9** (5 eq) followed by addition of aqueous NH₄Cl after 10 min gave a mixture containing an 89:11 ratio of β -amino esters **36**:**37** and less than 3% of oligomeric products, with chromatography furnishing β -amino esters (3*S*,4*R*, α *S*)-**36** and (3*R*,4*R*, α *S*)-**37** in 68% and 9% yield respectively and in >98% d.e. in each case (Scheme 7).

Having optimised the conjugate addition of homochiral lithium amide 9 to ester 18, the corresponding conjugate additions to dihydrothiophene and dihydropyrrole esters 25 and 21 were investigated. Addition of thiophene ester 25 to lithium amide (S)-9 gave, after addition of aqueous NH₄Cl, a mixture which contained the expected C(3)-epimeric β -amino esters (3*R*,4*R*, α S)-43 and (3*S*,4*R*, α S)-44 in a 69:31 ratio, with subsequent purification giving an inseparable 69:31



Scheme 6 Reagents and conditions: (i). lithium (S)-N-benzyl-N-a-methylbenzylamide 9, THF, -78 °C then $NH_4Cl_{(aq)}$, -78 °C to rt; (ii). KO'Bu, 'BuOH, Δ , 3 h.



Fig. 3 Chem 3D representation of the X-ray crystal structure of (3S,4R,aS)-40 (some H omitted for clarity).

mixture of $(3R,4R,\alpha S)$ -43 and $(3S,4R,\alpha S)$ -44 in 52% yield, and the deconjugated ester 32 in 29% yield. The purified 69:31 mixture of $(3R,4R,\alpha S)$ -43 and $(3S,4R,\alpha S)$ -44 was subjected to equilibrating conditions, furnishing the thermodynamic *trans*-diastereoisomer $(3S,4R,\alpha S)$ -44 in >98% d.e. The use of the hindered proton source 2,6-di-*tert*-butyl phenol in the



Fig. 4 Proposed mechanism for the synthesis of diamino ester 39.



Scheme 7 Reagents and conditions: (i). lithium (S)-N-benzyl-N- α -methylbenzylamide 9 (5 eq), THF, -78 °C, 10 min; (ii). NH₄- $Cl_{(aq)}$, -78 °C to rt.

reaction, which has previously been used to give high selectivity upon protonation of carbocyclic β -amino enolates,¹¹ did not improve the diastereoisomeric ratio, giving an identical product distribution. Conjugate addition of lithium amide (*S*)-**9** to dihydropyrrole ester **21** gave a mixture containing (3*R*,4*R*,*aS*)-**45** in >98% d.e., with chromatography giving (3*R*,4*R*,*aS*)-**45** in 46% isolated yield, plus the deconjugated ester **34** and the aromatic pyrrole **35** in 22% and 8% yield respectively. Attempted epimerisation of *cis*-**45** to the corresponding *trans*- β -amino ester proved unsuccessful, with the relative (3*R*,4*R*,*aS*)-configuration within **45** being assigned by analogy with that obtained from the addition of lithium dibenzylamide to **21**, and by its conversion into the known *cis*-4-amino-3-carboxylic acid (*vide infra*) (Scheme 8).

Deprotection: synthesis of *cis*-(3*R*,4*R*)- and *trans*-(3*S*,4*R*)-4-pyrrolidine-3-carboxylic acid and *cis*-(3*S*,4*R*)- and *trans*-(3*R*,4*R*)-4-tetrahydrofuran-3-carboxylic acid

With a range of heterocyclic β -amino esters prepared diastereoselectively, deprotection to the corresponding carboxylic acids was investigated. Attempted *N*-deprotection of the *trans*-dihydrothiophene β -amino ester **44** using a variety of palladium-based hydrogenolytic methods only returned predominantly starting material, while treatment with HBr²¹ led to complete decomposition. However, palladium-mediated hydrogenolytic *N*-debenzylation of *cis*-dihydrofuran β -amino ester (3*S*,4*R*, α *S*)-**36** gave the corresponding primary β -amino



Scheme 8 Reagents and conditions: (i). lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide 9 (1.6 eq), THF, -78 °C then NH₄Cl_(aq), -78 °C to rt; (ii). KO'Bu, 'BuOH, Δ , 3 h.

ester (3*S*,4*R*)-46, with subsequent treatment with TFA and purification by ion exchange chromatography furnishing acid (3*S*,4*R*)-47 in good yield (69% over two steps) and >98% d.e. and >97% e.e.²² Similarly, *N*-deprotection of *trans*-dihydrofuran β-amino ester (3*R*,4*R*,α*S*)-37, isolated either as the minor reaction component from the lithium amide conjugate addition reaction, or *via* quantitative epimerisation of *cis*-(3*S*,4*R*,α*S*)-36, gave the corresponding β-amino ester (3*R*,4*R*)-48, with ester hydrolysis and ion exchange chromatography giving (3*R*,4*R*)-49 in >98% d.e. and >97% e.e.²² (Scheme 9).



Scheme 9 Reagents and conditions: (i). $Pd(OH)_2$ on C, MeOH, H_2 (5 atm); (ii). TFA, rt then Dowex 50WX8-200; (iii). KO'Bu, 'BuOH, Δ , 3 h.

Similarly, palladium-mediated *N*-debenzylation of *cis*dihydropyrrole β -amino ester (3*R*,4*R*, α *S*)-**45** gave primary β -amino ester (3*R*,4*R*)-**50** in which the two amines are differentially protected, with subsequent treatment with TFA achieving concomitant ester and *N*-Boc deprotection, giving, after purification by ion exchange chromatography, (3*R*,4*R*)-4 in >98% d.e. (Scheme 10).

As this methodology only allows the asymmetric synthesis of the *cis*-stereoisomer of 4-pyrrolidine-3-carboxylic acid, an alternative strategy was investigated that would be amenable to the stereoselective synthesis of both *cis*- and *trans*-stereoisomers. It was envisaged that *N*-benzyl protection of the dihydropyrrole α,β -unsaturated ester would minimise γ -deprotonation, while being readily deprotected by hydrogenolysis. Thus, *tert*-butyl-*N*-benzyl-2,5-dihydropyrrole-3-carboxylate **52** was prepared in 93% isolated yield *via* 1,3-dipolar cycloaddition,



Scheme 10 Reagents and conditions: (i). $Pd(OH)_2$ on C, MeOH, H_2 (5 atm); (ii). TFA, rt then Dowex 50WX8-200.

following the method of Sakurai and Achiwa,²³ through treatment of commercially available *N*-(methoxymethyl)-*N*-(trimethylsilylmethyl)benzylamine **51** with a catalytic amount of acetic acid and *in situ* reaction with *tert*-butyl propiolate. Conjugate addition of lithium amide (*S*)-**9** to *N*-benzyl-**52** gave *cis*-(3*R*,4*R*, α *S*)-**53** in >98% d.e., with purification furnishing *cis*-(3*R*,4*R*, α *S*)-**53** in 79% isolated yield as a single stereoisomer (>98% d.e.). Subsequent treatment with KO'Bu in 'BuOH allowed quantitative conversion of **53** to the thermodynamic epimer *trans*-(3*S*,4*R*, α *S*)-**54** (Scheme 11).



Scheme 11 Reagents and conditions: (i). CH_3CO_2H , DCM, 0 °C then *tert*-butyl propiolate, 0 °C to rt, 4 h; (ii). lithium (*S*)-*N*-benzyl-*N*- α -methylbenzylamide, THF, -78 °C then NH₄Cl_(aq), -78 °C to rt; (iii). KO'Bu, 'BuOH, Δ , 3 h.

With an efficient asymmetric route to N-protected cis- and trans-4-aminopyrrolidine carboxylic esters in hand, methodology for the global and differential protection of the pyrrolidine scaffolds was investigated. Treatment of cis-(3R,4R,aS)-53 under standard palladium-mediated hydrogenolysis conditions {Pd(OH)₂ on C, H₂ (5 atm), using either MeOH, or MeOH: AcOH: $H_2O(40:4:1)$ as solvent} cleaved selectively the pyrrolidine N-benzyl group, giving $(3R, 4R, \alpha S)$ -55 in 69% yield, while hydrogenolysis in the presence of Boc₂O (1.05 eq) allowed the isolation of N-Boc-(3R, 4R)-56 in 59% yield. Alternatively, global N-deprotection could be achieved using the method of Nakayama et al.,24 with palladium-mediated hydrogenolysis of $(3R,4R,\alpha S)$ -53 in a mixture of EtOH, water and HCl (conc) effecting simultaneous cleavage of all three N-benzyl groups. Hydrolysis of the in situ made ethyl ester and purification by ion exchange chromatography gave cis-(3R,4R)-56 in >98% d.e. and 71% yield. Application of this global deprotection protocol to the trans-isomer $(3S, 4R, \alpha S)$ -54 gave trans-(3S, 4R)-7 as its hydrochloride salt in >98% d.e. and 68% overall yield (Scheme 12).

In conclusion, the conjugate addition of achiral and homochiral lithium amides to a range of dihydrofuran, *N*-protected dihydropyrrole and dihydrothiophene α,β -unsaturated esters has



Scheme 12 Reagents and conditions: (i). $Pd(OH)_2$ on C, MeOH, H_2 (5 atm) or $Pd(OH)_2$ on C, $MeOH:AcOH:H_2O$ (40:4:1), H_2 (5 atm); (ii). $Pd(OH)_2$ on C, Boc_2O (1.05 eq), MeOH, H_2 (5 atm); (iii). $Pd(OH)_2$ on C, EtOH, H_2O , HCl (conc), H_2 (5 atm) then 1 M $HCl_{(aq)}$, Δ , 4 h then Dowex 50WX8-200; (iv). $Pd(OH)_2$ on C, EtOH, H_2O , HCl (conc), H_2 (5 atm) then 1 M $HCl_{(aq)}$, Δ , 4 h.

been investigated. The diastereoselective conjugate addition of lithium (S)-N-benzyl-N- α -methylbenzylamide has been successfully applied to the first asymmetric syntheses of cis-(3S,4R)and trans-(3R,4R)-4-aminotetrahydrofuran-3-carboxylic acids in 26% and 25% overall yield, and to a new and straightforward synthesis of *cis*- and *trans*-4-aminopyrrolidine carboxylic acids. The four step synthesis of cis-(3R,4R)-4-aminopyrrolidine-3carboxylic acid 4 in 50% overall yield compares favourably with the only previously reported asymmetric synthesis by Aggarwal et al.,8 (nine steps, 15% overall yield). Similarly, the five step approach to trans-(3R,4S)-4-aminopyrrolidine-3-carboxylic acid hydrochloride 7 in 52% overall yield is more efficient than the ten step route (12% overall yield) towards (3S,4R)-7 detailed by Naito et al.9 As both enantiomers of lithium N-benzyl-N- α -methylbenzylamide are readily available in homochiral form, this methodology should be equally applicable to the asymmetric synthesis of any stereoisomer of these heterocyclic β -amino acids. The further application of the conjugate addition of homochiral lithium amides towards a variety of heterocyclic natural products is currently underway in our laboratory.

Experimental

General experimental

All reactions were carried out under nitrogen or argon using standard vacuum line techniques and glassware that was flame dried and cooled under nitrogen. THF was distilled from sodium/benzophenone ketyl; *n*-butyllithium was used as a solution in hexanes and was titrated against diphenylacetic acid prior to use. All other reagents were used as supplied without further purification. Flash column chromatography was performed on silica gel (Kieselgel 60). T.I.c was performed on Merck aluminium sheets coated with 0.2 mm silica gel 60 F₂₅₄. Plates were visualised either by UV light (254 nm), iodine, ammonium molybdate (7% solution in ethanol) or potassium permanganate (1% in 2% aqueous acetic acid, containing 7%

potassium carbonate). Infra red spectra were recorded as thin films or KBr discs using a Perkin-Elmer PARAGON 1000 FT-IR spectrometer, with selected peaks reported in cm⁻¹. ¹H and ¹³C NMR spectra were recorded on Varian Gemini 200 (1H 200 MHz, 13C 50 MHz), Bruker DPX-200 (1H 200 MHz, ¹³C 50 MHz), Bruker DPX-400 (¹H 400 MHz, ¹³C 100 MHz), or Bruker AM-500 (1H 500 MHz, 13C 125 MHz) spectrometers. Chemical shifts ($\delta_{\rm H}$) are reported in parts per million (p.p.m.) and are referenced to the residual solvent peak, with coupling constants (J) measured in Hertz. Low resolution mass spectra (m/z) were recorded on either a VG Masslab 20-250 instrument (CI, NH₃) or Platform instrument (APCI). MALDI spectra were recorded on a Micromass MALDI TOF SPEC 2E spectrometer. Major peaks are listed with intensities quoted as percentages of the base peak. Accurate mass measurements were recorded on a VG Autospec and a Waters 2790-Micromass LCT electrospray ionisation mass spectrometer operating at a resolution of 5000 full width half height. Positive ion spectra were calibrated relative to PEG with tetraoctylammonium bromide as the internal lock mass. Negative ion spectra were calibrated relative to poly-DLalanine with leucine enkephalin as the internal lock mass. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter, using a path length of 10 cm, in spectroscopic grade solvents (Aldrich), with concentrations (c) given in g per 100 cm^3 , solvent and temperature as recorded. Melting points were recorded on a Gallenkamp Hot Stage apparatus and are uncorrected.

General procedure 1: PPh₃/DIAD mediated elimination to furnish α,β-unsaturated acceptors

A solution of the β -hydroxy ester and PPh₃ (1.5 eq) in anhydrous THF was cooled to 0 °C prior to the dropwise addition of DIAD (1.3 eq). The reaction mixture was warmed to rt, whereat it was stirred overnight. The solvent was removed *in vacuo* and *n*-pentane (50 ml) added. After stirring at rt for 0.5 h, the precipitate was removed by filtration and the filtrate concentrated *in vacuo*. Additional *n*-pentane was added and the process repeated a further two times. Concentration *in vacuo* of the resultant filtrate furnished a yellow oil, which was passed through a silica gel plug (eluting 2% Et₂O/*n*-pentane) to give the requisite α , β -unsaturated acceptor.

General procedure 2: lithium amide conjugate additions

A solution of the amine in anhydrous THF under an inert atmosphere was cooled to -78 °C, prior to the slow addition of *n*-butyllithium (titrated before use, 1 eq). The resultant pink solution was stirred for 1 h at this temperature before the requisite α,β -unsaturated acceptor as a solution in anhydrous THF was added dropwise via syringe. The resulting mixture was stirred for 3 h at -78 °C after which time the reaction was quenched by addition of either (a) a precooled solution of 2,6di-tert-butylphenol in anhydrous THF; or (b) NH₄Cl (aq, sat) solution. The resultant mixture was kept at -78 °C for 0.5 h and then allowed to warm to rt over 1 h. NH₄Cl (aq, sat) solution was added and the mixture diluted with Et₂O. The organic layer was separated and the aqueous layer extracted with $Et_2O(3\times)$. The combined organic extracts were dried over MgSO₄, filtered and concentrated in vacuo to give the crude products. The individual products were purified as described.

General procedure 3: epimerisation of lithium amide adducts

To a solution of the substrate in *tert*-butanol was added a catalytic quantity of potassium *tert*-butoxide (≈ 20 mg). The resultant mixture was heated at reflux for 3 h then allowed to cool. The reaction was quenched by addition of NH₄Cl (aq, sat) and the mixture diluted with Et₂O. The organic layer was separated and the aqueous layer extracted with Et₂O (3×). The combined organic extracts were dried over MgSO₄, filtered and concentrated *in vacuo* to give the crude product. The individual products were purified as described.

General procedure 4: hydrogenolysis of lithium amide adducts using Pearlman's catalyst

A solution of the substrate in MeOH was placed in a Fischer– Porter bottle. The vessel was pump-filled five times with nitrogen prior to charging with $Pd(OH)_2(20\%)$ wt on carbon, 20% by mass of substrate used). The reaction mixture was stirred rapidly at rt overnight, after which time the solution was filtered through a pad of Celite[®], washed through with MeOH and concentrated *in vacuo* to give the crude product. The individual products were purified as described.

General procedure 5: TFA cleavage furnishing free amino acids

TFA was added to a solution of the crude β -amino ester at rt and stirred for 16 h. Concentration *in vacuo* gave an oil, which was dissolved in MeOH (2 ml) and HCl in Et₂O (sat, 2 ml). Concentration *in vacuo* gave a pale brown solid, which was partitioned between Et₂O (4 ml) and H₂O (4 ml). The aqueous phase was separated and concentrated to a quarter of its volume and chromatographed using Dowex[®] 50WX8-200 resin to give the free amino acid.

Preparation of tert-butyl 4-oxotetrahydrofuran-3-carboxylate 16

In a modification of the literature procedure,²⁵ sodium hydride (60% suspension in mineral oil, 8.0 g, 0.20 mol) was first prepared for use by careful washing with *n*-pentane $(3 \times$ 50 ml) and discarding the supernatant. The liberated sodium hydride was suspended in Et₂O (150 ml) and ethyl glycolate (18.02 g, 0.20 mol) added neat, dropwise. The reaction mixture was stirred for an additional 0.5 h after the evolution of H₂ gas had ceased. The Et₂O was removed *in vacuo* to give the sodium derivative as a white powder. tert-Butyl acrylate (35.2 ml, 0.24 mol) in anhydrous DMSO (150 ml) was added to this derivative in one lot, while the reaction flask was kept immersed in an ice bath. After 15 min, the cooling bath was removed and the reaction mixture stirred for an additional 0.5 h at rt, filtered, poured into ice-cold H₂SO₄ (5% aq, 100 ml) and extracted with Et₂O (3×150 ml). The combined organic layers were washed with brine (aq, sat, 200 ml), dried (MgSO₄), filtered and concentrated in vacuo to give the crude product as an oil. Purification by flash chromatography on silica gel ($10 \rightarrow 15\%$ EtOAc/*n*-pentane) gave 16 (25.3 g, 68%) as a white crystalline solid; m.p. 38–40 °C; elemental analysis, found C, 58.4; H, 7.4%; C₉H₁₄O₄ requires C, 58.1; H, 7.6%; v_{max} (CHCl₃) 3028s, 2983s, 2936m, 2883m, 1774s, 1724s, 1370s, 1330s, 1157s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.46 (9H, s, OC(CH₃)₃), 3.39 (1H, t, J 8.3, C(3)H), 3.92 and 3.99 (2 × 1H, AB system, J_{AB} 16.9, C(5) H_2), 4.38 and 4.46 (2 × 1H, AB system, J_{AB} 8.0, C(2) H_2); δ_C (100 MHz, CDCl₃) 27.9 (OC(CH_3)₃), 54.2 (C(3)), 69.6 (*C*(2)), 70.6 (*C*(5)), 82.8 (O*C*(CH₃)₃), 165.7 (*C*O₂^{*i*}Bu), 207.9 (C(4)O); m/z (GCMS, CI⁺) 204 (MH⁺, 100), 187 (MH⁺, 10), 148 (MNH₄⁺ – C₄H₈, 65%); HRMS, found 187.0968; C₉H₁₅O₄ (MH⁺) requires 187.0970.

Preparation of *tert*-butyl 4-hydroxytetrahydrofuran-3carboxylate 17

To a solution of β -keto ester **16** (20.0 g, 0.107 mol) in 'PrOH (300 ml) at 0 °C was added portionwise NaBH₄ (1.62 g, 42.8 mmol). After 2 h, NaBH₄ (1.62 g, 42.8 mmol) was again added portionwise and the temperature allowed to rise to rt. NaBH₄ (0.81 g, 21.4 mmol) was added again and stirring continued for 0.5 h. The reaction mixture was diluted with Et₂O (200 ml) and poured into brine (aq, sat, 200 ml). The aqueous layer was separated and extracted with Et₂O (2 × 200 ml). The combined organic extracts were washed with NaHCO₃ (aq, sat, 2 × 100 ml) and brine (aq, sat, 100 ml), then dried (MgSO₄), filtered and concentrated *in vacuo* to furnish crude **17** (17.8 g, 89%) as a colourless oil. This material was used without purification, although for the purposes of analysis a small quantity was subjected to flash chromatography on silica

gel $(5 \rightarrow 20 \rightarrow 50\% \text{ Et}_2\text{O}/n\text{-pentane})$ to give 17, a colourless oil, as a 1:1 mixture of diastereoisomers; v_{max} (film) 3430br s, 2978s, 2936m, 1728s, 1369s, 1158s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.43 and 1.45 $(2 \times 9H, 2 \times s, 2 \times OC(CH_3)_3), 2.93$ (1H, m, C(3)H diastereoisomer A), 3.06 (1H, br s, OH) overlays 3.04 (1H, m, C(3)H diastereoisomer B), 3.44 (1H, br s, OH), 3.72 (1H, dd, J 9.6, 2.5, $C(5)H_A B$, 3.81–3.89 (4H, m, $C(2)H_A A$, $C(5)H_2 A$ and $C(5)H_B$ B), 4.02 (2H, m, C(2)*H*₂ B), 4.15 (1H, app t, *J* 7.0, C(2)*H*_B A), 4.54–4.58 (2H, m, C(4)*H* A and C(4)*H* B); $\delta_{\rm C}$ (100 MHz, CDCl₃) $28.0 (2 \times OC(CH_3)_3), 50.3 (C(3) \text{ diastereoisomer B}), 54.5 (C(3))$ diastereoisomer A), 67.7 (C(2) B), 69.4 (C(2) A), 72.1 (C(4) A), 74.7 (C(4) B), 74.9 (C(5) B), 75.3 (C(5) A), 81.5 and 82.2 $(2 \times OC(CH_3)_3)$, 170.5 and 171.5 $(2 \times CO_2 Bu)$; m/z (GCMS, CI⁺) 206 (MNH₄⁺, 100), 189 (MH⁺, 40), 150 (MNH₄⁺ - C₄H₈, 90), 133 (MH⁺ – C_4H_8 , 15%); HRMS, found 189.1129; $C_9H_{17}O_4$ (MH⁺) requires 189.1127.

Preparation of tert-butyl 2,5-dihydrofuran-3-carboxylate 18

The β-hydroxy ester **17** (15.0 g, 79.8 mmol) in THF (50 ml) was treated with PPh₃ (31.5 g, 0.12 mol) and DIAD (20.4 ml, 0.104 mol) in accordance with *general procedure 1*, giving **18** (12.4 g, 91%) as a volatile, colourless liquid; v_{max} (film) 2980s, 2935m, 2876m, 1714s, 1647m, 1370s, 1284s, 1166s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.48 (9H, s, OC(CH₃)₃), 4.72–4.79 (4H, m, C(2)H₂ and C(5)H₂), 6.71 (1H, t, *J* 1.9, C(4)H); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 74.1 and 76.1 (*C*(2) and *C*(5)), 81.2 (OC(CH₃)₃), 134.6 (*C*(2)), 136.7 (*C*(3)), 161.8 (CO₂'Bu); *mlz* (GCMS, CI⁺) 188 (MNH₄⁺, 100), 171 (MH⁺, 55), 132 (MNH₄⁺ – C₄H₈, 60), 115 (MH⁺ – C₄H₈, 30%); HRMS, found 171.1020; C₉H₁₅O₃ (MH⁺) requires 171.1021.

Preparation of *tert*-butyl 4-oxotetrahydrothiophene-3carboxylate 22 and *tert*-butyl 4-hydroxy-2,5-dihydrothiophene-3carboxylate 23

A solution of ethyl thioglycolate (3.20 ml, 29.2 mmol) and tertbutyl acrylate (4.28 ml, 29.2 mmol) in THF (30 ml) was cooled to 0 °C. KO'Bu (3.81 g, 32.1 mmol) was added portionwise and the mixture allowed to warm to rt whereat it was stirred for 3 d. The crude reaction mixture was poured into ice-cold H₂SO₄ (5% aq, 30 ml) and extracted with Et_2O (3 × 30 ml). The combined organic phases were dried (MgSO₄), filtered and concentrated in vacuo. Purification by flash chromatography on silica gel (5% Et₂O/*n*-pentane) gave 22 and 23 (4.78 g, 81%), a 46:54 mixture of tautomers, as a white crystalline solid; m.p. 42–44 °C; v_{max} (CHCl₃) 3030m, 3006m, 2983s, 2935m, 1749s, 1724s, 1663s, 1395s, 1370s, 1334s, 1280s, 1247s, 1149s, 1110s, 842m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.48 (7.65H, s, OC(CH₃)₃ keto), 1.50 (9H, s, $OC(CH_3)_3$ enol), 3.18 (0.85H, dd, J 11.3, 7.5, C(2)H_A keto), 3.32 (1.7H, d, J 12.5, C(5)H₂ keto), 3.35 (0.85H, dd, J 11.3, 9.0, C(2)H_B keto), 3.39 (0.85H, dd, J 9.0, 7.5, C(3)H keto), 3.72 (2H, t, J 3.3, $C(2)H_2$ enol), 3.79 (2H, t, J 3.3, $C(5)H_2$ enol), 11.15 (1H, br s, OH enol); $\delta_{\rm C}$ (100 MHz, CDCl₃) 27.9 and 28.2 (2 × OC(CH₃)₃), 29.1 (C(2) keto), 31.8 (C(2) enol), 36.1 (C(5) enol), 37.4 (C(5) keto), 56.4 (C(3) keto), 82.0 and 82.8 ($2 \times OC(CH_3)_3$), 100.5 (C(3) enol), 166.9 and 169.0 $(2 \times CO_2^{t}Bu)$, 171.4 (C(4) enol), 206.2 (C(4)O keto); m/z (GCMS, CI⁺) 220 (MNH₄⁺, 35), 203 $(MH^+, 30)$, 164 $(MN_4^+ - C_4H_8, 40)$, 102 (100%); HRMS, found 203.0735; C₉H₁₅O₃S (MH⁺) requires 203.0742.

Preparation of *tert*-butyl 4-hydroxytetrahydrothiophene-3-carboxylate 24

To a solution of the keto–enol mixture **22** and **23** (2.45 g, 12.1 mmol) in 'PrOH (40 ml) at 0 °C was added portionwise NaBH₄ (246 mg, 6.50 mmol). After 2 h, NaBH₄ (246 mg, 6.50 mmol) was again added portionwise and the temperature allowed to rise to rt. NaBH₄ (123 mg, 3.25 mmol) was added again and stirring continued for 0.5 h. The reaction mixture was diluted with Et₂O (50 ml) and poured into brine (aq, sat, 50 ml). The aqueous layer was separated and extracted with

 Et_2O (2 × 50 ml). The combined organic extracts were washed with NaHCO₃ (aq, sat, 2×50 ml) and brine (aq, sat, 50 ml), dried (MgSO₄), filtered and concentrated in vacuo to furnish the crude product 24 (2.37 g, 96%), a colourless oil, as a 1:1 mixture of diastereoisomers. This material was used without purification, although for the purposes of analysis a small quantity was subjected to flash chromatography on silica gel $(5 \rightarrow 10 \rightarrow 20\%)$ Et_2O/n -pentane) to give the separable diastereoisomers: diastereoisomer A (first to elute) was obtained as a colourless oil; v_{max} (film) 3432br s, 2978s, 2937m, 1725s, 1155s, 1023m, 847m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.48 (9H, s, OC(CH₃)₃), 2.92 and 3.06 $(2 \times 1H, ABX \text{ system}, J_{AB} 11.7, J_{AX} 3.7, J_{BX} 1.7, C(5)H_2), 2.88-$ 2.91 (1H, m, C(3)H), 3.03 (1H, app t, J 10.4, C(2)H_A), 3.17 (1H, app t, J 10.7, C(2)H_B), 3.24 (1H, d, J 0.8, OH), 4.68–4.73 (1H, m, C(4)H); δ_C (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 29.7 (C(2)), 38.4 (C(5)), 53.7 (C(3)), 74.8 (C(4)), 82.2 (OC(CH₃)₃), 171.5 (CO₂^{*t*}Bu); *m*/*z* (GCMS, CI⁺) 222 (MNH₄⁺, 15), 205 (MH⁺, 100), 166 (MNH₄⁺-C₄H₈, 60), 149 (MH⁺-C₄H₈, 10%); HRMS, found 205.0902; C₉H₁₇O₃S (MH⁺) requires 205.0898; diastereoisomer B (second to elute) was obtained as a white, waxy solid; m.p. 38-40 °C; v_{max} (CHCl₃) 3507br m, 3008m, 2983m, 2941m, 1708s, 1152s, 1030m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.45 (9H, s, OC(CH₃)₃), 2.81 (1H, m, C(5)H_A), 2.92–2.97 (2H, m, C(3)H and OH), 3.03-3.07 (3H, m, C(2) H_2 and C(5) H_B), 4.56 (1H, m, C(4)H); δ_C (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 29.4 (C(2)), 36.6 (C(5)), 54.5 (C(3)), 76.2 (C(4)), 81.9 (OC(CH₃)₃), 171.4 (CO₂/Bu); m/z (GCMS, CI⁺) 222 (MNH₄⁺, 65), 205 (MH⁺, 100), 166 (MNH₄⁺ -C₄H₈, 70), 149 (MH⁺ – C₄H₈, 30%); HRMS, found 205.0892; C₉H₁₇O₃S (MH⁺) requires 205.0898.

Preparation of tert-butyl 2,5-dihydrothiophene-3-carboxylate 25

The β-hydroxy ester **24** (1.77 g, 8.65 mmol) in THF (50 ml) was treated with PPh₃ (3.40 g, 13.0 mmol) and DIAD (2.22 ml, 11.25 mmol) in accordance with *general procedure 1*, giving **25** (1.51 g, 94%) as a volatile, colourless liquid; v_{max} (film) 2977s, 2929m, 1708s, 1646m, 1368m, 1343m, 1273s, 1166s, 1079m, 857m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.48 (9H, s, OC(CH₃)₃), 3.88–3.96 (4H, m, C(2)H₂ and C(4)H₂), 6.78 (1H, m, C(4)H); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 37.2 and 38.8 (*C*(2) and *C*(5)), 81.2 (OC(CH₃)₃), 137.1 (*C*(3)), 139.5 (*C*(4)), 163.2 (CO₂'Bu); *mlz* (GCMS, CI⁺) 204 (MNH₄⁺, 15), 187 (MH₄⁺, 100%); HRMS, found 187.0791; C₉H₁₅O₂S (MH⁺) requires 187.0793.

Preparation of *tert*-butyl *N-tert*-butyloxycarbonyl-4oxopyrrolidine-3-carboxylate potassium salt 19

tert-Butyl*N-tert*-butyloxycarbonyl-4-oxopyrrolidine-3-carboxylate potassium salt **19** was prepared according to the literature procedure.⁴ Thus, a solution of Boc-glycine methyl ester (22.8 g, 0.112 mol) and *tert*-butyl acrylate (16.4 ml, 0.112 mol) in THF (200 ml) was cooled to 0 °C, and KO'Bu (14.6 g, 0.123 mol) added portionwise. The mixture was warmed to rt, whereat it was stirred for 24 h. Hexane was added to the thick mixture and the solid collected by filtration and washed with Et₂O (2 × 50 ml). The potassium salt **19** (27.9 g, 77%) was obtained as a pale yellow powder; m.p. 118–120 °C (decomposes); *m/z* (GCMS, CI⁺) 303 (MNH₄⁺, 100), 286 (MH⁺, 45), 247 (MNH₄⁺ – C₄H₈, 60), 228 (MH⁺ – C₄H₈, 25%).

Preparation of *tert*-butyl *N-tert*-butyloxycarbonyl-4hydroxypyrrolidine-3-carboxylate 20

The potassium salt **19** (15.5 g, 50 mmol), NaBH₃CN (3.46 g, 55 mmol) and bromocresol green (pH 3.8 yellow, pH 5.4 blue) were dissolved in MeOH (200 ml). HCl (1 M, aq) was added dropwise while stirring to maintain pH 3–4. After 15–20 min, the pH change slowed. The mixture was stirred for an additional 1 h, during which time HCl (1 M, aq) was added occasionally to maintain pH 3–4. Water (20 ml) was added and the mixture extracted with Et₂O (3 × 100 ml). The combined organic layers were washed with NaHCO₃ (aq, sat, 100 ml), brine (aq,

sat, 100 ml), then dried (MgSO₄), filtered and concentrated in vacuo to furnish the crude product as a waxy solid (13.6 g, 95%). This material was used without purification, although for the purposes of analysis a small quantity was subjected to flash chromatography on silica gel $(5\rightarrow 20\% \text{ EtOAc/n-pentane})$ to give 20, a 60:40 mixture of diastereoisomers, as a white waxy solid; m.p. 120-122 °C; v_{max} (CHCl₃) 3446br s, 3008s, 2982s, 2935m, 1690br s, 1417s, 1369s, 1246m, 1155s; $\delta_{\rm H}$ (400 MHz, CDCl₃, rotameric at 298 K) 1.48 and 1.50 (30H, $2 \times s$, $4 \times$ $OC(CH_3)_3$ minor diastereoisomer + major diastereoisomer) 3.11 (1H, ddd, J 14.3, 8.6, 4.1, C(3)H major), 3.27 (0.67H, m, C(3)H minor), 3.33 (0.67H, m, C(5)H_A minor), 3.42 (1H, m, C(5)H_A major), 3.46 (1H, dd, J 12.4, 3.8, C(5)H_B major), 3.53 (1H, dd, J10.6, 8.7, C(2)H_A major) overlays 3.51–3.54 (1.33H, m, $C(2)H_2$ minor), 3.65 (1H, app t, J 10.5, $C(2)H_B$ major) overlays 3.63 (0.67H, m, C(5)H_B minor), 4.49 (0.67H, m, C(4)H minor) overlays 4.51 (1H, m, C(4)H major); $\delta_{\rm C}$ (100 MHz, CDCl₃, rotameric at 298 K) 28.7/28.8 and 29.0/29.2 (4 × OC(CH₃)₃ minor diastereoisomer + major diastereoisomer), 46.4 and 46.7 (C(2) major and C(2) minor), 51.0 and 51.7 (C(3) major and C(3) minor), 55.8 and 56.1 (C(5) major and C(5) minor), 72.2 and 72.9 (C(4) major and C(4) minor), 81.4/81.5 and 82.7/ 82.9 (4 × OC(CH₃)₃), 156.8 and 156.9 (2 × NCO₂^tBu), 171.4 and 173.0 (2 × C(3) CO_2 'Bu); m/z (GCMS, CI⁺) 288 (MH⁺, 5), 232 (MH⁺ - C₄H₈, 10), 188 (100%); HRMS, found 288.1806; C₁₄H₂₆NO₅ (MH⁺) requires 288.1811.

Preparation of *tert*-butyl *N-tert*-butyloxycarbonyl-2,5dihydropyrrole-3-carboxylate 21

The β-hydroxy ester **20** (12.5 g, 43.5 mmol) in THF (300 ml) was treated with PPh₃ (17.1 g, 65.3 mmol) and DIAD (11.1 ml, 56.6 mmol) in accordance with *general procedure 1*, giving **21** (10.9 g, 93%) as a white solid; m.p. 54–56 °C; v_{max} (CHCl₃) 3026m, 3009s, 2982s, 2933m, 1701s, 1693s, 1412s, 1394s, 1300s, 1163s, 1095s; $\delta_{\rm H}$ (400 MHz, CDCl₃, rotameric at 298 K) 1.47/1.48 and 1.49/1.51 (18H, 4 × s, OC(CH₃)₃), 4.23 and 4.28 (2 × 2H, 2 × m, C(2)H₂ and C(5)H₂), 6.60 and 6.63 (2 × 1H, 2 × t, *J* 1.9, C(4)*H*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.0 and 28.4 (2 × OC(CH₃)₃), 51.8/52.0 and 53.5/53.7 (*C*(2) and *C*(5)), 79.7/53.7 and 81.2/81.3 (2 × OC(CH₃)₃), 133.7/133.8 (*C*(3)), 135.4/135.5 (*C*(4)), 153.9/154.1 (NCO₂'Bu), 162.1/162.2 (C(3)CO₂'Bu); *m/z* (GCMS, CI⁺) 270 (MH⁺, 15), 214 (MH⁺ - C₄H₈, 10), 170 (100%); HRMS, found 270.1713; C₁₄H₂₄NO₄ (MH⁺) requires 270.1705.

Preparation of *tert*-butyl (3SR,4RS)-4-(N,N-dibenzylamino)tetrahydrofuran-3-carboxylate 26, *tert*-butyl (3RS,4RS)-4-(N,Ndibenzylamino)tetrahydrofuran-3-carboxylate 27, di-*tert*-butyl (3SR, 4RS,3'SR,4'SR)-4-(N,N-dibenzylamino)tetrahydro-[3,3']-bifuranyl-3,4'-dicarboxylate 28 and di-*tert*-butyl (3SR,4RS,3'SR,4'RS)-4-(N,N-dibenzylamino)tetrahydro-[3,3']bifuranyl-3,4'-dicarboxylate 29

Following general procedure 2, n-BuLi (1.6 M, 2.94 ml, 4.70 mmol), dibenzylamine (927 mg, 4.70 mmol) in THF (25 ml) and 18 (500 mg, 2.94 mmol) in THF (5 ml) gave, after quenching with NH4Cl (aq, sat, 10 ml) the adducts 26:27:28:29 in a 17:26:52:5 ratio. Purification by flash chromatography on silica gel $(2 \rightarrow 4 \rightarrow 6 \rightarrow 8 \rightarrow 10 \rightarrow 20 \rightarrow 40\%)$ EtOAc: n-pentane) gave the products: (3RS,4RS)-27 (177 mg, 18%) as a white crystalline solid; m.p. 82-84 °C; elemental analysis, found C, 75.5; H, 8.1; N, 3.8%; C₂₃H₂₉NO₃ requires C, 75.2; H, 8.0; N, 3.8%; v_{max} (KBr) 2977m, 2932w, 2867w, 1723s, 1454m, 1368s, 1159s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.44 (9H, s, OC(CH₃)₃), 3.19 (1H, app td, J 8.0, 4.0, C(3)H), 3.56 and 3.75 $(2 \times 2H, AB \text{ system}, J_{AB} 14.0, N(CH_2Ph)_2), 3.78-3.84 (3H, m,$ $C(2)H_A$, C(4)H and $C(5)H_A$), 3.93 (1H, m, $C(5)H_B$), 4.18 (1H, app t, J 8.7, C(2) $H_{\rm B}$), 7.22–7.38 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 46.6 (C(3)), 54.7 (N(CH₂Ph)₂), 65.2 (C(4)), 71.0 (C(2) and C(5)), 81.0 (OC(CH₃)₃), 127.0 (p-Ph),

128.2 and 128.6 (o-, m-Ph), 139.3 (ipso-Ph), 173.0 (CO₂^tBu); m/z (APCI⁺) 368 (MH⁺, 40), 312 (MH⁺ - C₄H₈, 80), 311 (M⁺ - C₄H₈, 100%); HRMS, found 368.2221; C₂₃H₃₀NO₃ (MH⁺) requires 368.2226; (3SR,4RS)-26 (113 mg, 10%) as a white crystalline solid; m.p. 82–84 °C; v_{max} (KBr) 3011m, 2981m, 2888w, 1723s, 1369s, 1155s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.58 (9H, s, OC(CH₃)₃), 3.28 (1H, m, C(3)H), 3.64 and 3.80 (2 × 2H, AB system, J_{AB} 14.1, N(CH₂Ph)₂) overlays 3.64–3.70 (2H, m, C(4)*H* and C(5)*H*_A), 4.00 (1H, app t, *J* 8.9, C(2)*H*_A), 4.09 (1H, m, C(5)*H*_B), 4.27 (1H, dd, *J* 8.9, 6.1, C(2)*H*_B), 7.22–7.37 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.3 (OC(CH₃)₃), 49.3 (C(3)), 55.7 (N(CH₂Ph)₂), 63.8 (C(4)), 68.8 (C(5)), 69.8 (C(2)), 81.1 (OC(CH₃)₃), 126.9 (p-Ph), 128.1 and 128.9 (o-, m-Ph), 138.9 (ipso-Ph), 170.8 (CO2'Bu); m/z (APCI+) 368 (MH+, 15), 367 $(M^+, 15), 312 (MH^+ - C_4H_8, 85), 311 (MH^+ - C_4H_8, 100\%);$ HRMS, found 368.2225; C₂₃H₃₀NO₃ (MH⁺) requires 368.2226; **29** (363 mg, 46%) as a white crystalline solid; m.p. 86–88 °C; vmax (KBr) 3029m, 3011s, 2981s, 2934m, 1721s, 1369s, 1251s, 1153s, 1076m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.40 and 1.54 (2 × 9H, $2 \times s$, $2 \times OC(CH_3)_3$), 2.94 (1H, m, C(3')H), 3.11 (1H, m, C(4')H, 3.59 (1H, m, C(4)H) overlays 3.62 and 3.89 (2 × 2H, AB system, J_{AB} 14.5, N(CH₂Ph)₂), 3.74–3.78 (2H, m, C(2')H_A and C(5')*H*_A), 3.85–3.92 (3H, m, C(2)*H*_A, C(5)*H*_A and C(5')*H*_B), 3.96-4.03 (2H, m, C(2') H_B and C(5) H_B), 4.38 (1H, d, C(2) H_B), 7.19–7.39 (10H, m, Ph); δ_C (100 MHz, CDCl₃) 28.0 and 28.2 $(2 \times OC(CH_3)_3), 47.4 (C(3')), 49.3 (C(4')), 54.9 (N(CH_2Ph)_2),$ 60.8 (C(3)), 67.5 (C(4)), 68.5 (C(5)), 71.6 and 71.9 (C(2') and *C*(5')), 73.2 (*C*(2)), 81.2 and 82.1 (2 × O*C*(CH₃)₃), 127.0 (*p*-Ph), 128.2 and 128.7 (o-, m-Ph), 139.0 (ipso-Ph), 170.9 and 172.7 $(2 \times CO_2 Bu); m/z$ (APCI⁺) 538 (MH⁺, 100), 537 (M⁺, 100), 482 (MH⁺ - C₄H₈, 40), 481 (M⁺ - C₄H₈, 70%); HRMS, found 538.3161; C₃₂H₄₄NO₆ (MH⁺) requires 538.3169; **28** (31 mg, 4%) as a white crystalline solid; m.p. 80-82 °C; elemental analysis, found C, 71.7; H, 8.0; N, 8.1%; C₃₂H₄₃NO₆ requires C, 71.5; H, 8.1; N, 2.6%; v_{max} (KBr) 3010m, 2980s, 2936m, 1718s, 1369s, 1254m, 1152s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.45 and 1.47 (2 × 9H, $2 \times s$, $2 \times OC(CH_3)_3$), 2.14 (1H, m, C(3')H), 2.88 (1H, dt, J 11.1, 7.3, C(4')H), 3.56 and 3.92 (2 × 2H, AB system, J_{AB} 14.0, N(CH₂Ph)₂), 3.72 (1H, dd, J 8.6, 6.0, C(2')H_A), 3.80 (1H, dd, J 11.3, 7.6, C(5')H_A), 3.84–3.96 (4H, m, C(2)H_A, C(2')H_B, C(4)H and $C(5')H_B$, 3.99–4.07 (2H, m, $C(5)H_2$), 4.22 (1H, d, J 9.9, C(2) $H_{\rm B}$), 7.23–7.39 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 27.9 and 28.1 (2 × OC(CH₃)₃), 45.3 (C(3')), 48.5 (C(4')), 55.5 (N(CH₂Ph)₂), 56.7 (C(3)), 66.0 (C(5)), 68.2 (C(4)), 68.6 (C(5')), 71.2 (C(2)), 71.4 (C(2')), 81.4 and 82.4 (2 × O $C(CH_3)_3$), 127.1 (p-Ph), 128.0 and 128.8 (o-, m-Ph), 139.2 (ipso-Ph), 171.4 and $172.3 (2 \times CO_2{}^tBu); m/z (APCI^+) 538 (MH^+, 90), 537 (M^+, 100),$ $482 (MH^+ - C_4H_8, 10), 481 (M^+ - C_4H_8, 10\%); HRMS, found$ 538.3168; C₃₂H₄₄NO₆ (MH⁺) requires 538.3169.

Selective preparation of *tert*-butyl (*3RS*,*4RS*)-4-(*N*,*N*-dibenzylamino)tetrahydrofuran-3-carboxylate 27 and *tert*-butyl (*3SR*,*4RS*)-4-(*N*,*N*-dibenzylamino)tetrahydrofuran-3-carboxylate 26

Following general procedure 2, the lithium amide was prepared by addition of n-BuLi (2.5 M, 0.75 ml, 1.88 mmol) to a solution of dibenzylamine (371 mg, 1.88 mmol) in THF (10 ml) at -78 °C. A solution of 18 (200 mg, 1.18 mmol) in THF (5 ml) was added dropwise, via syringe pump, over 2 h. After addition was complete the reaction was stirred for a further 0.5 h, then quenched by addition of NH₄Cl (aq, sat, 10 ml). Following warming to rt, the aqueous phase was separated and extracted with Et₂O (3×20 ml). The combined organic layers were dried (MgSO₄), filtered and concentrated in vacuo to give (3RS,4RS)-27 and (3SR,4RS)-26 as a 42:58 mixture of diastereoisomers. Purification by flash chromatography on silica gel $(2 \rightarrow 4 \rightarrow 6 \rightarrow 8 \rightarrow 10\% \text{ EtOAc}: n$ pentane) gave the products (3RS,4RS)-27 (132 mg, 31%) and (3SR,4RS)-26 (171 mg, 40%), with spectroscopic data identical to that reported above.

Selective preparation of di-*tert*-butyl (3SR,4RS,3'SR,4'SR)-4-(N,N-dibenzylamino)tetrahydro-[3,3']-bifuranyl-3,4'dicarboxylate 28 and di-*tert*-butyl (3SR,4RS,3'SR,4'RS)-4-(N,N-dibenzylamino)tetrahydro-[3,3']-bifuranyl-3,4'dicarboxylate 29

Following general procedure 2, the lithium amide was prepared by addition of *n*-BuLi (2.5 M, 0.24 ml, 0.59 mmol) to a solution of dibenzylamine (116 mg, 0.59 mmol) in THF (5 ml) at -78 °C. A solution of **18** (200 mg, 1.18 mmol) in THF (5 ml) was added dropwise and the reaction mixture stirred at the reduced temperature for 5 h, then quenched by addition of NH₄Cl (aq, sat, 10 ml). Following warming to rt, the aqueous phase was separated and extracted with Et₂O (3 × 20 ml). The combined organic layers were dried (MgSO₄), filtered and concentrated *in vacuo* to give **29** and **28** as a 93:7 mixture of diastereoisomers. Purification by flash chromatography on silica gel (2 \rightarrow 10 \rightarrow 20 \rightarrow 40% EtOAc:*n*-pentane) gave the products **29** (203 mg, 71%) and **28** (21 mg, 7%), with spectroscopic data identical to that reported above.

Preparation of (*3RS*,4*RS*)-4-(*N*,*N*-dibenzylamino)tetrahydrofuran-3-carboxylate 7 *via* epimerisation

Following general procedure 3, a solution of (3SR,4RS)-26 (50 mg, 0.14 mmol) in 'BuOH (5 ml) was treated with KO'Bu and heated at reflux for 3 h. Purification by flash chromatography on silica gel (10% EtOAc:*n*-pentane) gave (3RS,4RS)-27 (50 mg, quantitative) in >98% d.e., with spectroscopic data identical to that reported above.

Preparation of di-*tert*-butyl (3SR,4RS,3'SR,4'RS)-4-(N,Ndibenzylamino)tetrahydro-[3,3']-bifuranyl-3,4'-dicarboxylate 29 *via* epimerisation

Following general procedure 3, a solution of **28** (50 mg, 0.09 mmol) in 'BuOH (5 ml) was treated with KO'Bu and heated at reflux for 3 h. Purification by flash chromatography on silica gel (30% EtOAc:*n*-pentane) gave **29** (49 mg, quantitative) in >98% d.e., with spectroscopic data identical to that reported above.

Preparation of *tert*-butyl (3*RS*,4*RS*)-4-(*N*,*N*-dibenzylamino)tetrahydrothiophene-3-carboxylate 30 and *tert*-butyl (3*SR*,4*RS*)-4-(*N*,*N*-dibenzylamino)tetrahydrothiophene-3-carboxylate 31

Following general procedure 2, n-BuLi (1.5 M, 0.57 ml, 0.86 mmol), dibenzylamine (170 mg, 0.86 mmol) in THF (5 ml) and 25 (100 mg, 0.54 mmol) in THF (2 ml) gave, after quenching with NH₄Cl (aq, sat, 10 ml), the adducts (3RS, 4RS)-30 and (3SR,4RS)-31 in a 54:46 ratio. Purification by flash chromatography on silica gel (2% Et₂O:n-pentane) gave (3RS,4RS)-30 and (3SR,4RS)-31 (99 mg, 49%), an inseparable mixture of diastereoisomers, as a colourless oil. The mixture of (3RS,4RS)-30 and (3SR,4RS)-31 (99 mg, 0.26 mmol) was redissolved in tert-butanol and epimerised under thermodynamic conditions in accordance with general procedure 3. Purification by flash chromatography on silica gel (2% Et₂O:*n*-pentane) gave (3SR, 4RS)-31 (98 mg, quantitative) in >98% d.e. as a colourless oil; v_{max} (film) 2978s, 2931s, 2856m, 1711s, 1394m, 1368s, 1278s, 1257s, 1163s, 1106s, 839m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.51 (9H, s, OC(CH₃)₃), 2.88 (2H, m, C(2)H₂), 2.91–2.93 (2H, m, C(5) H_2), 3.13 (1H, m, C(3)H), 3.49 and 3.57 (2 × 2H, AB system, J_{AB} 13.6, N(CH₂Ph)₂), 3.84 (1H, m, C(4)H), 7.23–7.36 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 30.8 (C(2)), 32.1 (C(5)), 51.9 (C(3)), 54.1 (N(CH₂Ph)₂), 55.7 (C(4)), 80.8 (OC(CH₃)₃), 127.0, 128.1, 128.4 (o-, m-, p-Ph), 138.7 (ipso-Ph), 171.1 (CO₂^tBu); m/z (APCI⁺) 384 (MH⁺, 20), 328 (MH⁺ - C₄H₈, 100%); HRMS, found 384.1995; C₂₃H₂₉NO₂S (MH⁺) requires 384.1997. Also isolated was the deconjugated adduct tert-butyl 2,3-dihydrothiophene-3-carboxylate 32 as a colourless oil (25 mg, 25%); v_{max} (film) 2952s, 2857m, 2844m, 1716s, 1631m, 1436s, 1297s, 1267s, 1197s, 1090s, 741m; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.47 (9H, s, OC(CH₃)₃), 3.36 (1H, dd, J 11.5, 10.1, C(2) H_A), 3.59 (1H, dd, J 11.5, 8.4, C(2) H_B), 3.86 (1H, m, C(3)H), 5.61 (1H, dd, J 5.9, 2.8, C(4)H), 6.25 (1H, dd, J 5.9, 2.5, C(5)H); δ_C (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 33.1 (C(2)), 53.9 (C(3)), 81.4 (OC(CH₃)₃), 120.7 (C(4)), 128.7 (C(5)), 171.2 (CO₂'Bu); m/z (GCMS, CI⁺) 204 (MNH₄⁺, 15), 187 (MH⁺, 100%); HRMS, found 187.0789; C₉H₁₅O₂S (MH⁺) requires 187.0793.

Preparation of *tert*-butyl *N*-*tert*-butyloxycarbonyl-(*3RS*,4*RS*)-4-(*N*,*N*-dibenzylamino)tetrahydropyrrole-3-carboxylate 33

Following general procedure 2, n-BuLi (1.6 M, 0.74 ml, 1.19 mmol), dibenzylamine (234 mg, 1.19 mmol) in THF (8 ml) and 21 (200 mg, 0.74 mmol) in THF (2 ml) gave, after quenching with NH₄Cl (aq, sat, 10 ml), (3RS,4RS)-33 in >98% d.e. Purification by flash chromatography on silica gel $(5 \rightarrow 10 \rightarrow 20\%$ EtOAc/*n*-pentane) gave (3RS, 4RS)-33 (141 mg, 41%) as a pale vellow crystalline solid; m.p. 61–63 °C; v_{max} (CHCl₃) 3009m, 2980m, 2933m, 1722s, 1687s, 1411s, 1368s, 1145s, 778s, 741s; $\delta_{\rm H}$ (400 MHz, CDCl₃, rotameric at 298 K) 1.44/1.47 and 1.53/1.56 (18H, $4 \times s$, $2 \times OC(CH_3)_3$), 3.24 (1H, m, C(3)H), 3.37 (1H, dd, J 10.1, 6.7, C(5)H_A), 3.41-3.52 (1.5H, m, C(2) H_A and C(4)H), 3.58 and 3.75 (2 × 1H, AB system, JAB 13.6, N(CH2Ph)2) overlays 3.58 (0.5H, m, C(4)H), 3.65 and 3.74 (2 × 1H, AB system, J_{AB} 13.9, N(CH₂Ph)₂) overlays 3.72-3.79 (2H, m, C(2)H_B and C(5)H_B), 7.22-7.34 (10H, m, *Ph*); $\delta_{\rm H}$ (400 MHz, d₈-toluene, 363 K) 1.45 and 1.46 (2 × 9H, 2 × s, 2 × OC(CH₃)₃), 2.90 (1H, m, C(3)H), 3.28 (1H, dd, J11.0, 7.5, C(2)*H*_A), 3.32 (2H, br, C(5)*H*₂), 3.64 (4H, br, N(C*H*₂Ph)₂), 3.81-3.86 (2H, br, C(2)H_B and C(4)H), 7.01-7.27 (10H, m, *Ph*); $\delta_{\rm C}$ (100 MHz, CDCl₃, rotameric at 298 K) 28.2/28.3 and $28.4/28.5 (2 \times OC(CH_3)_3)$, 46.5 and 47.1 (C(5)), 47.5 (C(3)), 47.9 and 48.2 (C(2)), 55.0 and 55.4 (N(CH₂Ph)₂), 61.1 and 62.9 (C(4)), 79.3/79.5 and 81.0/81.1 $(2 \times OC(CH_3)_3)$, 127.0, 127.1, 127.9, 128.1, 128.8 and 129.0 (o-, m-, p-Ph), 138.5/138.6 (ipso-Ph), 154.2/154.4 (NCO₂^tBu), 170.7/171.2 (C(3)CO₂^tBu); $\delta_{\rm C}$ (100 MHz, d_8 -toluene, 363 K) 28.3 and 28.6 (2 × OC(*C*H₃)₃), 48.2 (C(4), N(CH₂Ph)₂, C(5) and C(2)), 56.0 (C(3)), 78.7 and 80.5 (2 × OC(CH₃)₃), 127.1, 127.7, 127.8, 128.0, 128.1, 128.3, 128.5, 128.9, 129.0 and 129.4 (o-, m-, p-Ph), 137.4 and 139.4 (*ipso-Ph*), 154.1 (NCO₂^{*i*}Bu), 178.0 (C(3)CO₂^{*i*}Bu); *m*/*z* (APCI⁺) 467 (MH⁺, 100), 466 (M⁺, 95), 411 (MH⁺ - C₄H₈, 40), 410 $(M^+ - C_4H_8, 75\%)$; HRMS, found 467.2908; $C_{28}H_{39}N_2O_4$ (MH⁺) requires 467.2910. Also isolated was the deconjugated adduct tert-butyl N-tert-butyloxycarbonyl-2,3-dihydropyrrole-3-carboxylate 34 (44 mg, 22%) as a white solid; m.p. 58-60 °C; v_{max} (KBr) 3026w, 3009m, 2982s, 2933m, 1726s, 1697s, 1413s, 1369s, 1139s, 888m; $\delta_{\rm H}$ (400 MHz, d₈-toluene, rotameric at 298 K) 1.32/1.33 and 1.39/1.41 (18H, $4 \times s$, $2 \times OC(CH_3)_3$), 3.33-3.37 (0.5H, m, C(3)H), 3.40-3.44 (0.5H, m, C(3)H) overlays 3.47 (0.5H, t, J 11.3, C(2)H_A), 3.66 (0.5H, t, J 11.6, $C(2)H_A$, 4.21 (0.5H, dd, J 9.4, 4.7, $C(2)H_B$), 4.31 (0.5H, dd, J 11.7, 6.6, C(2)*H*_B), 4.86 (1H, dd, *J* 4.2, 2.6, C(4)*H*), 6.47 (0.5H, m, C(5)H), 6.77 (0.5H, m, C(5)H); $\delta_{\rm H}$ (500 MHz, d₈-toluene, 363 K) 1.33 and 1.40 (2 × 9H, 2 × s, 2 × OC(CH_3)₃), 3.42 (1H, m, C(3)*H*), 3.63 (1H, t, *J* 11.5, C(2)*H*_A), 4.16 (1H, dd, *J* 11.5, 6.5, C(2)H_B), 4.86 (1H, dd, J 4.0, 3.0, C(4)H), 6.59 (1H, br, C(5)H; δ_{C} (100 MHz, d₈-toluene, rotameric at 298 K) 28.1/28.3 and $28.4/28.6 (2 \times OC(CH_3)_3)$, 47.7 (C(2)), 48.3/49.5 (C(3)), 80.2 and 80.8 $(2 \times OC(CH_3)_3)$, 105.6/105.7 (C(4)), 131.8/12.1 $(C(5)), 151.8/152.8 (NCO_2 Bu), 171.2/171.3 (C(3)CO_2 Bu); \delta_C$ (125 MHz, d_8 -toluene, 363 K) 28.0 and 28.3 (2 × OC(*C*H₃)₃), 47.8 (C(2)), 48.7 (C(3)), 79.8 and 80.4 ($2 \times OC(CH_3)_3$), 105.3 (C(4)), 131.8 (C(5)), 151.2 $(NCO_2 Bu)$, 170.8 $(C(3)CO_2 Bu)$; *m*/*z* (GCMS, CI⁺) 270 (MH⁺, 10%), 214 (MH⁺ - C₄H₈, 20), 170 (100%); HRMS, found 270.1703; C14H24NO4 (MH+) requires 270.1705; and the aromatised adduct tert-butyl N-Boc pyrrole-1,3-dicarboxylate **35** (22 mg, 11%) as a white crystalline solid; m.p. 70–72 °C; v_{max} (KBr) 3028m, 3009m, 2983s, 2935m, 1749s, 1703s, 1561s, 1496s, 1389s, 1369s, 1289s, 1254s, 1147s, 1100s,

1067s, 975s, 963s, 848s, 831s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.55 and 1.60 (2 × 9H, 2 × s, (2 × OC(CH₃)₃), 6.55 (1H, dd, *J* 3.3, 1.6, C(4)*H*), 7.17 (1H, dd, *J* 3.3, 2.2, C(5)*H*), 7.74 (1H, dd, *J* 2.2, 1.6, C(2)*H*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 27.9 and 28.2 (2 × OC(CH₃)₃), 80.3 and 84.8 (2 × OC(CH₃)₃), 112.0 (*C*(4)), 120.3 (*C*(3)), 121.2 (*C*(5)), 124.3 (*C*(2)), 148.2 (NCO₂^{*i*}Bu), 163.5 (C(3)CO₂^{*i*}Bu); *m*/*z* (GCMS, CI⁺) 268 (MH⁺, 15), 212 (MH⁺ - C₄H₈, 100%); HRMS, found 268.1541; C₁₄H₂₂NO₄ (MH⁺) requires 268.1549.

Preparation of *tert*-butyl (3R,4R, αS)-4-(N-benzyl-N- α -methylbenzylamino)tetrahydrofuran-3-carboxylate 37, *tert*-butyl (3S,4R, αS)-4-(N-benzyl-N- α -methylbenzylamino)tetrahydrofuran-3-carboxylate 36, di-*tert*-butyl (3S,4R,3'S,4'S, αS)-4-(N-benzyl-N- α -methylbenzylamino)tetrahydro-[3,3']-bifuranyl-3,4'-dicarboxylate 38 and *tert*-butyl (3S,4R,3'S,4'R, αS , $\alpha'S$)-4-(N-benzyl-N- α -methylbenzylamino)-3-[4'-(N-benzyl-N- α methylbenzylamino)-2'-oxotetrahydrofuran-3'-ylmethyl]tetrahydrofuran-3-carboxylate 39

Following general procedure 2, n-BuLi (2.5 M, 3.76 ml, 9.40 mmol), (S)-N-benzyl-N-α-methylbenzylamine (1.99 g, 9.40 mmol) in THF (50 ml) and 18 (1.00 g, 5.88 mmol) in THF (10 ml) gave, after quenching with NH₄Cl (aq, sat, 20 ml), the adducts 37:36:38:39 in a 4:47:2:47 ratio. Purification by flash chromatography on silica gel $(2 \rightarrow 4 \rightarrow 6 \rightarrow 8 \rightarrow 10 \rightarrow 20 \rightarrow 40\%)$ EtOAc: *n*-pentane) gave the products: $(3R, 4R, \alpha S)$ -37 as a white crystalline solid (67 mg, 3%); m.p. 68–70 °C; $[a]_{D}^{24}$ +67.7 (c1.5, CHCl₃); v_{max} (KBr) 3011m, 2980s, 2935m, 2872m, 1718s, 1369s, 1149s; δ_H (400 MHz, CDCl₃) 1.30 (3H, d, J 6.9, C(α)Me), 1.39 (9H, s, OC(CH₃)₃), 2.94 (1H, app q, J 8.0, C(3)H), 3.65 (1H, app t, J 8.5, C(2) H_A), 3.77 (1H, m, C(5) H_A), 3.85–3.91 (2H, m, C(4)Hand $C(5)H_B$ overlays 3.85 (2H, app s, NCH₂Ph), 4.06 (1H, app t, J 8.5, $C(2)H_B$, 4.15 (1H, q, J 6.8, $C(\alpha)H$), 7.21–7.47 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 16.3 (C(α)Me), 28.0 (OC(CH₃)₃), 50.1 (C(3) and NCH₂Ph), 58.0 (C(a)H), 63.3 (C(4)), 70.7 (C(2) and C(5)), 80.7 (OC(CH₃)₃), 126.8 (p-Ph), 127.7, 128.1, 128.3 and 128.3 (o, m-Ph), 141.2 and 143.7 (ipso-Ph), 172.6 (CO₂^tBu); m/z (APCI⁺) 382 (MH⁺, 30), 381 (M⁺, 50), 326 (MH⁺ - C₄H₈, 15), 222 (30), 221 (100%); HRMS, found 382.2381; C₂₄H₃₂NO₃ (MH⁺) requires 382.2382; $(3S, 4R, \alpha S)$ -36 as a white crystalline solid (694 mg, 31%); m.p. 84–86 °C; [*a*]_D²⁴ –48.0 (*c* 1.5, CHCl₃); elemental analysis, found C, 75.4; H, 8.1; N, 8.2%; C₂₄H₃₁NO₃ requires C, 75.6; H, 8.2; N, 3.8%; v_{max} (KBr) 3029w, 3010m, 2980s, 2936w, 1721s, 1369s, 1153s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.40 (3H, d, J 6.9, C(α)Me), 1.54 (9H, s, OC(CH₃)₃), 3.16 (1H, td, J 7.7, 4.8, C(3)H), 3.55–3.64 (2H, m, C(4)H and C(5)H_A), 3.61 and 4.03 (2 × 1H, AB system, J_{AB} 14.8, NC H_2 Ph), 3.83 (1H, app t, J 7.8, $C(5)H_B$, 3.92 (1H, dd, J 8.8, 7.8, $C(2)H_A$), 4.11 (1H, dd, J $8.8, 4.8, C(2)H_B$, $4.23 (1H, q, J 6.9, C(\alpha)H), 7.23-7.40 (10H, m, m)$ *Ph*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 16.3 (C(α)*Me*), 28.2 (OC(*C*H₃)₃), 49.3 $(C(3)), 51.9 (NCH_2Ph), 57.7 (C(\alpha)H), 63.4 (C(4)), 69.7 (C(5)),$ 70.1 (C(2)), 81.0 (OC(CH₃)₃), 126.7 and 127.0 (p-Ph), 128.0, 128.1 and 128.2 (o-, m-Ph), 141.1 and 141.8 (ipso-Ph), 171.7 (CO2^tBu); m/z (APCI⁺) 382 (MH⁺, 70), 381 (M⁺, 60), 326 (MH⁺-C₄H₈, 65), 325 (M⁺-C₄H₈, 100), 222 (45), 221 (100%); HRMS, found 382.2372; C₂₄H₃₂NO₃ (MH⁺) requires 382.2382; 38 as a white crystalline solid (32 mg, 2%); m.p. 82–84 °C; $[a]_{D}^{24}$ –38.0 (c 0.25, CHCl₃); v_{max} (KBr) 3011m, 2980s, 2936m, 2889m, 1721s, 1369s, 1153s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.15 (3H, d, J 6.9, C(α)Me), 1.42 and 1.46 (2×9 H, $2 \times s$, $2 \times OC(CH_3)_3$), 2.81–2.90 (2H, m, C(3')H and C(4')H), 3.51 (1H, dd, J 7.4, 5.3, C(4)H), 3.60 (2H, app d, J 7.0, C(5')H₂), 3.66 (1H, dd, J 8.8, 6.3, C(2')H_A), 3.80 $(1H, d, J 9.8, C(2)H_A)$ overlays 3.82 $(1H, m, C(2')H_B)$, 3.87 and 4.00 (2 × 1H, AB system, J_{AB} 14.5, NC H_2 Ph), 3.95 (1H, dd, J9.8, 7.6, C(5) H_A), 4.07–4.12 (2H, m, C(α)H and C(5) H_B), 4.27 (1H, d, J 9.8, C(2) $H_{\rm B}$), 7.46–7.51 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 15.0 (C(α)*Me*), 28.0 and 28.2 (2 × OC(*C*H₃)₃), 47.6 (*C*(3')), 49.6 (C(4')), 51.4 (NCH₂Ph), 57.5 ($C(\alpha)$ H), 60.3 (C(3)) 64.8 (C(4)), 69.1 (*C*(5)), 71.1 (*C*(5')), 71.8 (*C*(2')), 73.9 (*C*(2)), 81.0 and 81.9 (2 × OC(CH₃)₃), 126.7 and 126.9 (*p*-Ph), 128.0, 128.2, 128.4 and 128.5 (o-, m-Ph), 140.4 and 143.9 (ipso-Ph), 170.4 and 172.8

 $(2 \times CO_2 Bu); m/z$ (APCI⁺) 552 (MH⁺, 100), 551 (M⁺, 95), 496 $(MH^+ - C_4H_8, 30), 495 (MH^+ - C_4H_8, 35\%); HRMS, found$ 552.3319; C₃₃H₄₆NO₆ (MH⁺) requires 552.3325; **39** as a white crystalline solid (283 mg, 14%); m.p. 122–124 °C; [a]_D²⁴ –34.8 (c 1.0, CHCl₃); v_{max} (KBr) 3028m, 3012m, 2976m, 1770s, 1721s, 1494m, 1451m, 1153s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 0.93 (3H, d, J 6.8, C(*α*)*Me*), 1.27 (3H, d, *J* 6.9, C(*α'*)*Me*), 1.44 (9H, s, OC(CH₃)₃), 1.87 (1H, dd, J14.6, 4.5, C(3)CH_AH_BC(3')), 2.30 (1H, dd, J14.6, 2.9, C(3)CH_AH_BC(3')), 2.39 (1H, m, C(3')H), 2.70-2.74 (2H, m, C(4)H and C(4')H, 3.30 and 3.50 (2 × 1H, AB system, J_{AB} 13.1, NCH₂Ph), 3.58 (1H, app d, J 9.6, C(5)H_A) overlays 3.62 (1H, dd, J 10.0, 7.5, C(5')H_A), 3.79–3.86 (4H, m, C(α)H, C(α')H and NCH₂Ph), 3.86–3.94 (2H, m, C(2)H_A and C(5')H_B), 4.23 (1H, app d, J 9.6, C(5)H_B), 4.36 (1H, d, J 10.0, C(2)H_B), 7.21-7.53 (20H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 12.3 (C(α')Me), 14.2 $(C(\alpha)Me)$, 28.2 $(OC(CH_3)_3)$, 31.5 $(C(3)CH_2C(3'))$, 38.7 (C(3')), 50.2 and 51.1 (2 × NCH₂Ph), 54.3 (C(3)), 55.3 and 56.6 ($C(\alpha)$ H and *C*(*a*')H), 56.7 and 62.3 (*C*(4) and *C*(4')), 68.2 (*C*(5')), 69.0 (C(2)), 76.0 (C(5)), 81.6 (OC(CH₃)₃), 126.4, 127.0, 127.5 and 127.8 (p-Ph), 128.0, 128.2, 128.3, 128.4, 128.6, 128.7, 128.9 and 129.1 (o-, m-Ph), 138.2, 139.4, 142.7 and 143.9 (ipso-Ph), 170.2 (CO₂^{*i*}Bu), 179.4 (C(2')O); *m*/*z* (APCI⁺) 689 (MH⁺, 50), 688 (M⁺, 100%); HRMS, found 689.3952; C44H53N2O5 (MH+) requires 689.3954.

A fifth product was also isolated from the column and was identified as being tert-butyl $(3S, 4R, \alpha S)$ -4-(N-benzyl-N-α-methylbenzylamino)-3-(2'-oxo-2',5'dihydrofuran-3'-ylmethyl)tetrahydrofuran-3-carboxylate 40. $(3S, 4R, \alpha S)$ -40 was obtained as clear crystals (238 mg, 17%); m.p. 110–112 °C; [a]_D²⁴ –47.8 (c 0.90, CHCl₃); elemental analysis, found C, 73.0; H, 7.6; N, 2.9%; C₂₉H₃₅NO₅ requires C, 72.9; H, 7.4; N, 2.9%; v_{max} (KBr) 3028m, 3012m, 2977m, 1757s, 1714s, 1369m, 1231m, 1149s, 1069s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.26 (3H, d, J 6.9, C(a)Me), 1.38 (9H, s, OC(CH₃)₃), 1.70 (1H, dd, J 15.8, 1.6, C(3)CH_AH_BC(3')), 2.70 (1H, dd, J 15.8, 1.2, C(3)CH_ACH_BC(3')), 3.39 (1H, app t, J 7.4, C(4)H), 3.66 (1H, d, J 9.2, C(2) H_A), 3.78 and 4.02 (2 × 1H, AB system, J_{AB} 14.1, NCH₂Ph), 3.92 (1H, dd, J 8.9, 7.3, C(5)H_A), 4.02 (1H, q, J 6.9, C(α)H), 4.10 (1H, dd, J 8.9, 7.7, C(5)H_B), 4.25 (1H, d, J 9.2, C(2)H_B), 4.65 (2H, app d, J 1.8, C(5')H₂), 6.83 (1H, t, J 1.5, C(4')H), 7.23–7.53 (10H, m, Ph); δ_C (100 MHz, CDCl₃) 12.7 (C(α)Me), 28.0 (OC(CH₃)₃), 31.7 (C(3)CH₂C(3')), 51.5 (NCH₂Ph), 56.2 and 56.7 (C(a)H and C(3)), 66.6 (C(4)), 67.2 (C(5)), 69.9 (C(5')), 73.8 (C(2)), 82.0 (OC(CH₃)₃), 127.1 (p-Ph), 128.1, 128.2, 128.3 and 128.9 (o-, m-Ph), 131.0 (C(3')), 139.9 and 143.4 (*ipso-Ph*), 145.9 (*C*(4')), 171.7 (*CO*₂^{*i*}Bu), 174.2 (*C*(2')O); m/z (APCI⁺) 478 (MH⁺, 80), 477 (M⁺, 100), 422 (MH⁺ - C₄H₈, 45), 421 (M⁺ – C₄H₈, 80%); HRMS, found 478.2593; C₂₉H₃₆NO₅ (MH⁺) requires 478.2593.

X-Ray crystal structure determination for 40

Data were collected using an Enraf-Nonius ĸ-CCD diffractometer with graphite monochromated Cu-Ka radiation using standard procedures at 190 K. The structure was solved by direct methods (SIR92), all non-hydrogen atoms were refined with anisotropic thermal parameters. Hydrogen atoms were added at idealised positions. The structure was refined using CRYSTALS.26 X-Ray crystal structure data for 40 [$C_{29}H_{35}NO_5$]: M = 477.60, orthorhombic, space group $P2_{1}2_{1}2_{1}$, a = 8.8594(1) Å, b = 10.0807(2) Å, c = 28.3944(5) Å, V = 2535.87(7) Å³, Z = 4, $\mu = 0.085$ mm⁻¹, colourless block, crystal dimensions = $0.2 \times 0.2 \times 0.2$ mm. A total of 3252 unique reflections were measured for $5^{\circ} < \theta < 27^{\circ}$ and 2713 reflections were used in the refinement. The final parameters were $wR_2 = 0.040$ and $R_1 = 0.035$ [$I > 2\sigma(I)$]. Crystallographic data (excluding structure factors) has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC-237687. See http://www.rsc.org/ suppdata/ob/b4/b407558g/ for crystallographic data in .cif format.

Selective preparation of *tert*-butyl $(3R,4R,\alpha S)$ -4-(*N*-benzyl-*N*- α -methylbenzylamino)tetrahydrofuran-3-carboxylate 37 and *tert*-butyl $(3S,4R,\alpha S)$ -4-(*N*-benzyl-*N*- α -methylbenzylamino)-tetrahydrofuran-3-carboxylate 36

Following general procedure 2, the lithium amide was prepared by addition of *n*-BuLi (2.5 M, 1.18 ml, 2.94 mmol) to a solution of (*S*)-*N*-benzyl-*N*- α -methylbenzylamine (623 mg, 2.94 mmol) in THF (10 ml) at -78 °C. A cooled (-78 °C) solution of **18** (100 mg, 0.59 mmol) in THF (2 ml) was added rapidly *via* cannula and the reaction mixture stirred at the reduced temperature for 10 min. The reaction was quenched by addition of NH₄Cl (aq, sat, 10 ml), warmed to rt and the aqueous phase separated and extracted with Et₂O (3 × 20 ml). The combined organic layers were dried (MgSO₄), filtered and concentrated *in vacuo* to give (3*R*,4*R*,*aS*)-**37**, (3*S*,4*R*,*aS*)-**36** and **39** in an 11:86:3 ratio. Purification by flash chromatography on silica gel (2 \rightarrow 4 \rightarrow 6 \rightarrow 8% EtOAc:*n*-pentane) gave the products (3*R*,4*R*,*aS*)-**37** (20 mg, 9%) and (3*S*,4*R*,*aS*)-**36** (153 mg, 68%), with spectroscopic data identical to that reported above.

Preparation of *tert*-butyl (3*R*,4*R*,α*S*)-4-(*N*-benzyl-*N*-α-methyl-benzylamino)tetrahydrofuran-3-carboxylate 37 *via* epimerisation

Following general procedure 3, a solution of $(3S,4R,\alpha S)$ -36 (65 mg, 0.17 mmol) in 'BuOH (5 ml) was treated with KO'Bu and heated at reflux for 3 h. Purification by flash chromatography on silica gel (10% EtOAc:*n*-pentane) gave (3*R*,4*R*,*\alpha S*)-37 (64 mg, quantitative) in >98% d.e., with spectroscopic data identical to that reported above.

Preparation of *tert*-butyl $(3R,4R,\alpha S)$ -4-(N-benzyl-N- α -methylbenzylamino)tetrahydrothiophene-3-carboxylate 43 and *tert*-butyl $(3S,4R,\alpha S)$ -4-(N-benzyl-N- α methylbenzylamino)tetrahydrothiophene-3-carboxylate 44

Following general procedure 2, n-BuLi (1.6 M, 2.70 ml, 4.32 mmol), (S)-N-benzyl-N- α -methylbenzylamine (916 mg, 4.32 mmol) in THF (30 ml) and 25 (500 mg, 2.70 mmol) in THF (5 ml) gave, after quenching with NH₄Cl (aq, sat, 20 ml), $(3R,4R,\alpha S)$ -43 and $(3S,4R,\alpha S)$ -44 in a 69:31 ratio. Purification by flash chromatography on silica gel (2% Et₂O: *n*-pentane) gave $(3R,4R,\alpha S)$ -43 and $(3S,4R,\alpha S)$ -44 (557 mg, 52%), an inseparable mixture of diastereoisomers, as a colourless oil; NMR data for $(3R,4R,\alpha S)$ -43 (assigned from the diastereoisomeric mixture): $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.42 (3H, d, J 6.9, C(α)Me), 1.49 (9H, s, OC(CH₃)₃), 2.73 (1H, dd, *J* 10.1, 6.2, C(5)*H*_A), 2.86 (2H, app d, J 13.3, C(2)H₂), 3.07 (1H, dd, J 11.3, 5.3, C(3)H), 3.14 (1H, app t, J 10.1, C(5) H_B), 3.51 (1H, m, C(4)H), 3.79 and 3.96 (2 × 1H, AB system, J_{AB} 14.9, NCH₂Ph), 4.20 (1H, q, J 6.9, C(α)H), 7.24–7.45 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, CDCl₃) 15.7 (C(α)Me), 28.0 (OC(CH₃)₃), 30.6 (C(2)), 31.8 (C(5)), 49.9 (C(3)), 51.7 (NCH₂Ph), 57.8 (C(α)H), 66.9 (C(4)), 81.0 (OC(CH₃)₃), 126.7, 127.0, 127.9 and 128.1 (o-, m-, p-Ph), 141.4 and 142.3 (ipso-Ph), 172.2 (CO₂^tBu). Also isolated was the deconjugated adduct tertbutyl 2,3-dihydrothiophene-3-carboxylate 32 as a colourless oil (140 mg, 28%), with spectroscopic data identical to that reported above.

The mixture of $(3R,4R,\alpha S)$ -**43** and $(3S,4R,\alpha S)$ -**44** (300 mg, 0.76 mmol) was re-dissolved in *tert*-butanol and epimerised under thermodynamic conditions in accordance with *general procedure 3*. Purification by flash chromatography on silica (2% Et₂O:*n*-pentane) gave $(3S,4R,\alpha S)$ -**44** (257 mg, 86%) as a colourless oil; $[a]_D^{24}$ +109.0 (*c* 0.60, CHCl₃); v_{max} (film) 3085m, 3061m, 3027s, 2975s, 2934s, 2874m, 1724s, 1453s, 1367s, 1279s, 1254s, 1150s, 1028s, 846s, 749s; δ_H (400 MHz, CDCl₃) 1.40 (3H, d, *J* 7.0, C(α)*Me*), 1.44 (9H, s, OC(CH₃)₃), 2.80 (2H, app d, *J* 8.7, C(2)*H*₂), 2.84–2.89 (2H, m, C(5)*H*₂), 2.98 (1H, app q, *J* 9.0, C(3)*H*), 3.74 and 3.94 (2 × 1H, AB system, J_{AB} 14.4, NC*H*₂Ph), 3.87 (1H, m, C(4)*H*), 3.98 (1H, q, *J* 7.0, C(α)*Me*), 28.0 (OC(CH₃)₃), 29.5 and 30.7 (*C*(2) and *C*(5)), 49.6 (NCH₂Ph), 52.1 (*C*(3)), 57.9 (*C*(α)H),

66.2 (*C*(4)), 80.8 (O*C*(CH₃)₃), 126.9, 128.0, 128.1 and 128.6 (*o*-, *m*-, *p*-Ph), 140.8 and 143.3 (*ipso*-Ph), 171.9 (*C*O₂'Bu); *m*/*z* (GCMS, CI⁺) 398 (MH⁺, 100), 342 (MH⁺ – C₄H₈, 10%); HRMS, found 398.2152; C₂₄H₃₁NO₂S (MH⁺) requires 398.2154.

Preparation of *tert*-butyl *N*-*tert*-butyloxycarbonyl-(3*R*,4*R*,α*S*)-4-(*N*-benzyl-*N*-α-methylbenzylamino)tetrahydropyrrole-3carboxylate 45

Following general procedure 2, n-BuLi (1.6 M, 3.72 ml, 5.94 mmol), (S)-N-benzyl-N- α -methylbenzylamine (1.26 g, 5.94 mmol) in THF (40 ml) and 21 (1.00 g, 3.72 mmol) in THF (10 ml) gave, after quenching with NH₄Cl (aq, sat, 40 ml), $(3R,4R,\alpha S)$ -45 in >98% d.e. Purification by flash chromatography on silica gel $(5 \rightarrow 10 \rightarrow 20\%$ EtOAc/*n*-pentane) gave $(3R, 4R, \alpha S)$ -45 (821 mg, 46%) as a colourless oil; $[a]_D^{25} - 18.5$ (c 1.8, CHCl₃); v_{max} (CHCl₃) 3008m, 2980s, 2934m, 1719s, 1687s, 1412s, 1368s, 1152s; $\delta_{\rm H}$ (400 MHz, d₈-toluene, rotameric at 298 K) 1.20 (1.5H, d, J 6.9, $C(\alpha)Me$, 1.29 (1.5H, d, J 6.9, $C(\alpha)Me$), 1.41/1.42 and 1.43/1.46 (18H, 4 × s, 2 × OC(CH₃)₃), 2.82 (0.5H, m, C(3)H), 2.87 (0.5H, m, C(3)H), 3.03–3.11 (1.5H, m, C(2) H_A and C(5) H_A), 3.17 and 3.84 (2 × 0.5H, AB system, J_{AB} 14.9, NCH₂Ph), 3.19 (0.5H, m, C(4)H), 3.27–3.32 (1H, m, C(2)H_B and C(4)H), 3.38 (0.5H, dd, J 10.5, 7.1, C(5) H_A), 3.52 and 3.74 (2 × 0.5H, AB system, J_{AB} 14.6, NC H_2 Ph), 3.61–3.66 (1H, m, C(2) H_B and C(5) H_B), 3.76 (0.5H, dd, J 11.6, 10.1, C(5)H_B), 4.16 (0.5H, q, J 6.9, C(a)H), 4.30 $(0.5H, q, J 6.9, C(\alpha)H)$, 7.06–7.38 (10H, m, Ph); $\delta_{\rm H}$ (500 MHz, d₈-toluene, 363 K) 1.28 (3H, d, J 6.5, C(α)Me), 1.40 and 1.43 $(2 \times 9H, 2 \times s, 2 \times OC(CH_3)_3), 2.83 (1H, m, C(3)H), 3.13 (1H, m)$ app dd, J11.5, 7.0, C(4)H), 3.36 (2H, m, C(2)H₂), 3.59-3.64 (2H, br, NC H_A H_BPh and C(5) H_A), 3.66 (1H, br, C(5) H_B), 3.88 (1H, d, J 14.5, NCH_A H_B Ph), 4.25 (1H, q, J 6.5, C(α)H), 7.15–7.33 (10H, m, Ph); $\delta_{\rm C}$ (100 MHz, d₈-toluene, rotameric at 298 K) 15.9/17.7 $(C(\alpha)Me)$, 28.4/28.5 and 28.8/28.9 $(2 \times OC(CH_3)_3)$, 47.9 and 48.0 (C(5) and C(3)), 48.3 and 48.4 (C(5) and C(2)), 49.2 (C(2) and C(3)), 52.3/52.5 (NCH₂Ph), 57.7/58.2 (C(a)H), 61.4/63.6 (C(4)), 78.8/79.0 and 80.6/80.7 $(2 \times OC(CH_3)_3)$, 127.3, 127.4, 127.5, 127.7, 128.1, 128.3, 128.6, 128.7, 128.8, 129.0, 129.2 and 129.5 (o-, m-, p-Ph), 141.8 and 142.5/142.9 (ipso-Ph), 154.2/154.6 $(NCO_2 Bu)$, 171.8/172.0 (C(3)CO_2 Bu); δ_C (125 MHz, d₈-toluene, 363 K) 28.4/28.5 and 28.7/28.9 (2 × OC(CH₃)₃), 48.2 and 48.7 $(C(2), C(3), C(4) \text{ and } C(5)), 52.5 (NCH₂Ph), 58.4 (C(<math>\alpha$)H), 78.8 and 80.6 $(2 \times OC(CH_3)_3)$, 127.1, 127.3, 127.9, 128.1, 128.3, 128.4, 128.5, 128.6, 128.8, 128.9, 129.0 and 129.2 (o-, m-, p-Ph), 141.3 and 141.9 (ipso-Ph), 154.3 (NCO₂^tBu), 171.7 $(C(3)CO_2^{t}Bu); m/z (APCI^{+}) 481 (MH^{+}, 100), 425 (MH^{+} - C_4H_8),$ 20%); HRMS, found 481.3166; C₂₉H₄₁N₂O₂ (MH⁺) requires 481.3168. Also isolated was the deconjugated adduct tert-butyl N-Boc-2,3-dihydropyrrole-3-carboxylate 34 (220 mg, 22%) as a white solid and the aromatised adduct tert-butyl N-Boc pyrrole-1,3-dicarboxylate 35 (79 mg, 8%) as a white crystalline solid, both with spectroscopic data identical to that reported above.

Preparation of *tert*-butyl (3*S*,4*R*)-4-aminotetrahydrofuran-3-carboxylate 46

Following general procedure 4, Pd(OH)₂ on C (60 mg) was added to a stirred degassed solution of (3S,4R,aS)-**36** (300 mg, 0.79 mmol) in MeOH (5 ml) and stirred under H₂ (5 atm) overnight. Filtration through Celite[®] and concentration *in vacuo* gave the crude β -amino ester. Although this material was used without purification, flash chromatography on silica gel (Et₂O) gave an analytical sample of (3S,4R)-**46** as a colourless oil; $[a]_D^{24}$ -3.2 (*c*1.0, CHCl₃); ν_{max} (CHCl₃) 2982m, 2877w, 1720s, 1370m, 1154s; δ_{H} (400 MHz, CDCl₃) 1.48 (9H, s, OC(CH₃)₃) overlays 1.50 (2H, br s, NH₂), 3.01 (1H, app q, *J* 7.7, C(3)*H*), 3.59 (1H, dd, *J* 8.9, 3.8, C(5)*H*_A), 3.76 (1H, br m, C(4)*H*), 3.93 (1H, td, *J* 8.7, C(2)*H*_B); δ_{C} (100 MHz, CDCl₃) 28.2 (OC(CH₃)₃), 50.9 (C(3)), 54.2 (C(4)), 68.2 (C(2)), 75.4 (C(5)), 81.3 (OC(CH₃)₃), 170.4 (CO₂'Bu); *m*/z (GCMS, CI⁺) 188

(MH⁺, 50), 132 (MH⁺–C₄H₈, 100%); HRMS, found 188.1280; C₉H₁₈NO₃ (MH⁺) requires 188.1287.

Preparation of *tert*-butyl (3*R*,4*R*)-4-aminotetrahydrofuran-3-carboxylate 48

Following general procedure 4. Pd(OH)₂ on C (50 mg) was added to a stirred degassed solution of $(3S, 4R, \alpha S)$ -37 (250 mg, 0.66 mmol) in MeOH (5 ml) and stirred under H_2 (5 atm) overnight. Filtration through Celite® and concentration in vacuo gave the crude β -amino ester. Although this material was used without purification, flash chromatography on silica gel (Et₂O) gave an analytical sample of (3R, 4R)-48 as a colourless oil; [a]_D²⁴ +61.9 (c 2.1, CHCl₃); v_{max} (CHCl₃) 2982m, 2872w, 1721s, 1370m, 1154s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.43 (9H, s, OC(CH₃)₃), 1.72 (2H, br s, NH₂), 2.66–2.71 (1H, m, C(3)H), 3.46 (1H, dd, J 8.8, 5.2, C(5)*H*_A), 3.72 (1H, app q, *J* 5.4, C(4)*H*), 3.89 (1H, dd, *J* 9.0, 6.8, C(2)*H*_A), 3.95 (1H, dd, *J* 8.8, 6.2, C(5)*H*_B), 4.10 (1H, app t, J 8.7, C(2)H_B); δ_C (100 MHz, CDCl₃) 28.0 (OC(CH₃)₃), 55.0 (C(3)), 56.1 (C(4)), 69.7 (C(2)), 75.3 (C(5)), 81.2 (OC(CH₃)₃), 171.8 (CO2'Bu); m/z (GCMS, CI+) 188 (MH+, 15), 132 (MH⁺ - C₄H₈, 100%); HRMS, found 188.1289; C₉H₁₈NO₃ (MH⁺) requires 188.1287.

Preparation of (3S,4R)-4-aminotetrahydrofuran-3-carboxylic acid 47

Following *general procedure 5*, TFA (5 ml) was added to a solution of crude β-amino ester (3*S*,4*R*)-**46** (80 mg, 0.43 mmol) at rt and stirred for 16 h. Purification using Dowex[®] 50X8-200 resin gave (3*S*,4*R*)-**47** (39 mg, 69% from **36**) as a white solid; m.p. 198–200 °C (decomposes); $[a]_D^{23}$ +19.3 (*c* 0.80, H₂O); v_{max} (KBr) 3482br s, 3081–2794br s, 2572m, 2182br s, 1620s, 1562s, 1408s, 1185s, 1093s, 901s, 777s; $\delta_{\rm H}$ (400 MHz, D₂O) 3.30 (1H, app qd, *J* 9.0, 2.4, C(3)*H*), 3.85–3.91 (3H, m, C(2)*H*_A and C(5)*H*₂), 3.93 (1H, m, C(4)*H*), 4.05 (1H, app t, *J* 9.2, C(2)*H*_B); $\delta_{\rm C}$ (100 MHz, D₂O) 47.3 (*C*(3)), 52.1 (*C*(4)), 69.2 (*C*(2)), 71.0 (*C*(5)), 176.8 (*C*O₂H); *m*/*z* (ESI⁺) 132 (MH⁺, 100%); HRMS, found 132.0656; C₅H₁₀NO₅ (MH⁺) requires 132.0661.

Preparation of (3*R*,4*R*)-4-aminotetrahydrofuran-3-carboxylic acid 49

Following general procedure 5, TFA (5 ml) was added to a solution of crude β-amino ester (3*R*,4*R*)-48 (75 mg, 0.40 mmol) at rt and stirred for 16 h. Purification using Dowex[®] 50X8-200 resin gave (3*R*,4*R*)-49 (35 mg, 66% from 37) as a white solid; m.p. 213–215 °C (decomposes); $[a]_D^{23}$ +17.8 (*c* 1.0, H₂O); v_{max} (KBr) 3462br s, 3021–2763br s, 2347–1931br s, 1623s, 1565s, 1396s, 1372s, 1077s, 901s, 856s; $\delta_{\rm H}$ (400 MHz, D₂O) 2.93–2.97 (1H, m, C(3)*H*), 3.71 (1H, app t, *J* 7.8, C(2)*H*_A), 3.80–3.88 (2H, m, C(5)*H*₂), 4.03 (1H, m, C(4)*H*), 4.17 (1H, app t, *J* 8.8, C(2)*H*_B); $\delta_{\rm C}$ (100 MHz, D₂O) 51.8 (*C*(3)), 54.8 (*C*(4)), 71.0 and 71.1 (*C*(2) and *C*(5)), 177.9 (*C*O₂H); *m*/*z* (ESI⁺) 132 (MH⁺, 100%); HRMS, found 132.0663; C₅H₁₀NO₃ (MH⁺) requires 132.0661.

Preparation of *tert*-butyl (3*R*,4*R*)-*N*-tert-butyloxycarbonyl-4aminopyrrolidine-3-carboxylate 50

Following general procedure 4, Pd(OH)₂ on C (120 mg) was added to a stirred degassed solution of $(3R,4R,\alpha S)$ -45 (600 mg, 1.25 mmol) in MeOH (8 ml) and stirred under H₂ (5 atm) overnight. Filtration through Celite[®] and concentration *in vacuo* gave the crude β -amino ester. Although this material was used without purification, flash chromatography on silica gel (Et₂O) gave an analytical sample of (3R,4R)-50 (211 mg, 59%) as a white solid; m.p. 55–57 °C; $[a]_D^{24}$ +4.7 (*c* 0.90, CHCl₃); ν_{max} (CHCl₃) 3005m, 2981s, 2935m, 1722s, 1689s, 1414s, 1368s, 1154s; δ_H (400 MHz, d₄-MeOH, rotameric at 298 K) 1.43/1.44 and 1.51 (18H, 3 × s, 2 × OC(CH₃)₃), 3.14 (1H, m, C(3)H), 3.30 (1H, dd, *J* 11.1, 2.6, C(5)H_A), 3.50–3.55 (2H, m, C(2)H_A and C(5)H_B),

3.64 (1H, m, C(2) H_B), 3.73 (1H, m, C(4)H); δ_C (100 MHz, d₄-MeOH, rotameric at 298 K) 28.6/28.7 and 29.2 (2 × OC(CH₃)₃), 47.3/47.5 (*C*(2)), 48.8/49.0 (*C*(3)), 53.0/53.8 (*C*(4)), 54.3/54.6 (*C*(5)), 81.5 and 83.3 (2 × OC(CH₃)₃), 156.7 (NCO₂'Bu), 172.3 (C(3)CO₂'Bu); *m/z* (GCMS, CI⁺) 287 (MH⁺, 60), 231 (MH⁺ - C₄H₈, 80), 175 (MH⁺ - C₈H₁₆, 100%); HRMS, found 287.1970; C₁₄H₂₇N₂O₄ (MH⁺) requires 287.1971.

Preparation of (3R,4R)-4-aminopyrrolidine-3-carboxylic acid 4

Following general procedure 5, TFA (5 ml) was added to a solution of crude β -amino ester (3*R*,4*R*)-**50** (200 mg, 0.70 mmol) at rt and stirred for 16 h. Purification using Dowex[®] 50X8-200 resin gave (3*R*,4*R*)-4 (65 mg, 71% from **45**) as a pale yellow solid; m.p. 164–166 °C (decomposes) (lit.⁸ 166–167 °C); [*a*]_D²⁵+17.1 (*c* 2.5, H₂O) {lit.⁸ [*a*]_D²²+25.9 (*c* 0.9, MeOH)}; ν_{max} (KBr) 3581–2372br s, 3420s, 1576s, 1410s, 1127m, 618m; $\delta_{\rm H}$ (400 MHz, D₂O) 3.05 (1H, br m, C(3)*H*) overlays 3.10 (1H, dd, *J* 11.9, 4.3, C(5)*H*_A), 3.32–3.43 (3H, m, C(2)*H*₂ and C(5)*H*_B), 3.77 (1H, m, C(4)*H*); $\delta_{\rm C}$ (100 MHz, D₂O) 46.6 (*C*(2)), 49.9 (*C*(3)), 50.9 (*C*(5)), 51.4 (*C*(4)), 176.6 (*C*O₂H); *m*/*z* (ESI⁺) 153 (MNa⁺, 70), 131 (MH⁺, 100%); HRMS, found 131.0818; C₅H₁₁N₂O₂ (MH⁺) requires 131.0821.

Preparation of tert-butyl propiolate

A solution of propiolic acid (25.0 g, 0.36 mol) in Et₂O (100 ml) was cooled to -78 °C and ≈ 100 ml of isobutylene (condensed by passing the gas into a conical flask held at -78 °C) added, followed by H₂SO₄ (98%, 1 drop). The reaction mixture was held at -78 °C for 4 h, then allowed to warm to rt overnight. The mixture was diluted with water (100 ml) and the aqueous layer separated and extracted with Et₂O (3 × 100 ml). The combined organic layers were dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash chromatography on silica gel (5% Et₂O/*n*-pentane) gave the title compound (33.6 g, 74%) as a crystalline solid; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.51 (9H, s, OC(CH₃)₃), 2.76 (1H, s, C=CH), with all other spectroscopic data identical to the commercially available material.²⁷

Preparation of *tert*-butyl *N*-benzyl-2,5-dihydropyrrole-3carboxylate 52

To a solution of N-(methoxymethyl)-N-(trimethylsilylmethyl)benzylamine 51²² (5.0 g, 19.8 mmol) in DCM (75 ml) at 0 °C was added dropwise a solution of *tert*-butyl propiolate (2.1 g, 16.5 mmol) in DCM (10 ml). Acetic acid (0.09 ml, 1.65 mmol) was added neat, dropwise, and the reaction mixture allowed to warm to rt, whereat it was stirred for 4 h. The reaction was neutralized by addition NaHCO₃ (aq, sat, 50 ml) and the organic phase separated, washed with brine (aq, sat, 50 ml) and then dried (MgSO₄), filtered and concentrated in vacuo. Purification by flash chromatography on silica gel (20% EtOAc: n-pentane)gave 52 (3.97 g, 93%) as a colourless oil; v_{max} (film) 3003m, 2976s, 2932m, 1706s, 1455m, 1367s, 1246s, 1166s, 762m, 731m, 701m; δ_H (400 MHz, CDCl₃) 1.48 (9H, s, OC(CH₃)₃), 3.61–3.68 (4H, m, C(2)H₂ and C(5)H₂), 3.82 (2H, s, NCH₂Ph), 6.66 (1H, m, C(4)H), 7.23–7.37 (5H, m, Ph); δ_C (100 MHz, CDCl₃) 28.1 (OC(CH₃)₃), 58.3 and 60.0 (C(2) and C(5)), 80.6 (OC(CH₃)₃), 127.2 (p-Ph), 128.4 and 128.6 (o-, m-Ph), 135.6 (C(3)), 138.1 (ipso-Ph), 138.9 (C(4)), 162.9 (CO₂ⁱBu); m/z (GCMS, CI⁺) 260 (MH⁺, 100%); HRMS, found 260.1648; C₁₆H₂₂NO₂ (MH⁺) requires 260.1651.

Preparation of *tert*-butyl *N*-benzyl (3*R*,4*R*,α*S*)-4-(*N*'-benzyl-*N*'*α*-methylbenzylamino)tetrahydropyrrole-3-carboxylate 53

Following general procedure 2, *n*-BuLi (2.5 M, 2.5 ml, 6.2 mmol), (*S*)-*N*-benzyl-*N*- α -methylbenzylamine (1.31 g, 6.2 mmol) in THF (50 ml) and **52** (1.00 g, 3.9 mmol) in THF (10 ml) gave, after quenching with NH₄Cl (aq, sat, 40 ml), (3*R*,4*R*, α *S*)-**53** in >98% d.e. Purification by flash chromatography on silica gel

 $(5 \rightarrow 10 \rightarrow 20\%$ EtOAc/*n*-pentane) gave $(3R, 4R, \alpha S)$ -53 (1.45 g, 79%) as a pale yellow solid; m.p. 66–68 °C; $[a]_D^{25}$ +32.0 (c 1.0, CHCl₃); v_{max} (KBr) 3061m, 3030m, 2977s, 2934s, 1724s, 1455s, 1368s, 1152s, 750m, 701s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.52 (3H, d, J 6.8, C(a)Me), 1.60 (9H, s, OC(CH₃)₃), 2.70 (1H, dd, J 15.3, 3.9, C(2) H_A), 2.77–2.83 (2H, m, C(2) H_B and C(5) H_A), 2.96 (1H, m, C(3)H), 3.09 and 4.07 (2 × 1H, AB system, J_{AB} 15.4, NCH₂Ph), 3.11 (1H, m, C(5)H_B), 3.67 (1H, m, C(4)H), 4.08 and 4.26 (2 × 1H, AB system, J_{AB} 14.1, N'CH₂Ph), 4.29 (1H, d, J 6.8, C(α)*H*), 7.12–7.32 (15H, m, *Ph*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 14.9 (C(α)Me), 28.2 (OC(CH₃)₃), 46.7 (C(2)), 51.3 (NCH₂Ph), 53.9 and 54.1 (C(3) and C(5)), 58.2 (C(a)H), 59.5 (N'CH₂Ph), 62.2 (C(4)), 82.9 (OC(CH₃)₃), 127.0, 127.1, 127.7, 128.0, 128.2, 128.4, 128.5, 128.6, 129.1, 129.3, 129.5, 129.8, 130.0, 130.5, 130.7 (o-, m-, p-Ph), 138.9 and 140.7 (ipso-Ph), 172.4 (CO₂^tBu); m/z (APCI⁺) 471 (MH⁺, 100), 470 (M⁺, 65), 415 (MH⁺ - C₄H₈, 15), 414 (M⁺ – C₄H₈, 20%); HRMS, found 471.2989; C₃₁H₃₉N₂O₂ (MH⁺) requires 471.3012.

Preparation of *tert*-butyl *N*-benzyl (3S,4R,aS)-4-(N'-benzyl-N'-a-methylbenzylamino)tetrahydropyrrole-3-carboxylate 54 *via* epimerisation

Following general procedure 3, a solution of $(3R,4R,\alpha S)$ -53 (700 mg, 1.49 mmol) in 'BuOH (10 ml) was treated with KO'Bu and heated at reflux for 3 h. Purification by flash chromatography on silica gel (20% EtOAc: *n*-pentane) gave $(3S, 4R, \alpha S)$ -54 (695 mg, quantitative) in >98% d.e. as a colourless oil; $[a]_D^{25}$ +51.6 (c 1.1, CHCl₃); v_{max} (film) 3028m, 2974s, 2928s, 1725s, 1453s, 1367s, 1150s, 747s, 699s; δ_H (400 MHz, CDCl₃) 1.31 (3H, d, J 6.8, $C(\alpha)Me$, 1.40 (9H, s, $OC(CH_3)_3$), 2.35 (1H, m, $C(2)H_A$), 2.51 (1H, app t, J9.1, C(5)H_A), 2.92–3.00 (3H, m, C(2)H_B, C(3)H and $C(5)H_B$, 3.57 and 3.63 (2 × 1H, AB system, J_{AB} 13.2, NC H_2 Ph), 3.81 (1H, m, C(4)H), 3.92 and 4.08 (2 × 1H, AB system, J_{AB} 15.3, N'CH₂Ph), 3.94 (1H, q, J 6.8, C(α)H), 7.21–7.52 (15H, m, *Ph*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 15.9 (C(α)*Me*), 28.0 (OC(*C*H₃)₃), 49.6 (C(3)), 50.3 (NCH₂Ph), 56.6 (C(2)), 57.3 (C(5)), 57.9 (C(α)H), 59.9 (N'CH₂Ph), 61.3 (C(4)), 80.2 (OC(CH₃)₃), 126.6, 126.8 and 126.9 (p-Ph), 127.2, 127.8, 128.0, 128.2, 128.4, 128.6 (o-, m-Ph), 138.9, 142.0, 144.3 (ipso-Ph), 173.6 (CO₂^tBu); m/z (ESI⁺) 471 $(MH^+, 100)$, 415 $(MH^+ - C_4H_8, 15\%)$; HRMS, found 471.3011; $C_{31}H_{39}N_2O_2$ (MH⁺) requires 471.3012.

Preparation of *tert*-butyl (3*R*,4*R*,α*S*)-4-(*N*-benzyl-*N*-α-methylbenzylamino)pyrrolidine-3-carboxylate 55

Following general procedure 4, Pd(OH)₂ on C (70 mg) was added to a stirred degassed solution of $(3R, 4R, \alpha S)$ -53 (344 mg, 0.73 mmol) in MeOH (5 ml) and stirred under H₂ (5 atm) overnight. Filtration through Celite® and concentration in vacuo gave the crude pyrrolidine as a yellow oil. Purification by flash chromatography on silica gel ($20 \rightarrow 40\%$ EtOAc/*n*-pentane) gave $(3R,4R,\alpha S)$ -55 (191 mg, 69%) as a clear oil; $[a]_{D}^{25}$ -35.7 (c 2.0, CHCl₃); *v*_{max} (film) 3421br w, 2978s, 2932s, 1725s, 1602w, 1454m, 1369s, 1153s, 910s, 744s, 701s; $\delta_{\rm H}$ (400 MHz, CDCl₃) 1.42 (3H, d, J 6.8, C(a)Me), 1.53 (9H, s, OC(CH₃)₃), 3.10-3.21 (3H, m, C(2)H₂ and C(5)H_A), 3.27 (1H, dd, J12.2, 6.9, C(5)H_B), 3.35 and 4.07 (2 × 1H, AB system, J_{AB} 15.4, NC H_2 Ph), 3.47–3.59 (2H, m, C(3)H and C(4)H), 4.24 (1H, q, J 6.8, C(α)H), 7.22–7.40 (10H, m, *Ph*); $\delta_{\rm C}$ (100 MHz, CDCl₃) 17.1 (C(α)*Me*), 28.1 (OC(*C*H₃)₃), 45.9 and 45.7 (C(2) and C(5)), 47.0 (C(3)), 51.7 (NCH₂Ph), 57.9 (C(α)H), 62.6 (C(4)), 83.0 (OC(CH₃)₃), 126.6 and 127.2 (p-Ph), 127.4, 127.6, 128.4 and 128.5 (o-, m-Ph), 140.2 (ipso-Ph), 170.4 (CO2'Bu); m/z (ESI+) 381 (MH+, 100%); HRMS, found 381.2546; C₂₄H₃₃N₂O₂ (MH⁺) requires 381.2542.

Alternative preparation of *tert*-butyl (3*R*,4*R*)-*N*-*tert*-butyloxycarbonyl-4-aminopyrrolidine-3-carboxylate 56

Following general procedure 4, $Pd(OH)_2$ on C (40 mg) was added to a stirred degassed solution of $(3R,4R,\alpha S)$ -53 (200 mg, 0.43 mmol) and di-*tert*-butyl dicarbonate (98 mg, 0.45 mmol)

in MeOH (5 ml) and stirred under H₂ (5 atm) overnight. Filtration through Celite® and concentration in vacuo gave the crude β -amino ester. Purification by flash chromatography on silica gel (Et₂O) gave an analytical sample of $(3R, 4R, \alpha S)$ -56 (69 mg, 56%) as a white solid, with spectroscopic data identical to that reported above.

Alternative preparation of (3R,4R)-4-aminopyrrolidine-3carboxylic acid 4

Following general procedure 4, Pd(OH)₂ on C (30 mg) was added to a stirred degassed solution of $(3R, 4R, \alpha S)$ -53 (150 mg, 0.43 mmol) in a mixture of EtOH (2 ml), H₂O (1 ml) and HCl (conc, 0.5 ml). The reaction mixture was stirred under H₂ (5 atm) overnight, then filtered through Celite® and concentrated in vacuo to give the crude (3R,4R) β -amino ester as a yellow oil (65 mg). This material was subsequently re-dissolved in HCl (1 M aq, 5 ml) and heated at 70 °C for 4 h. Following cooling to room temperature, the solution was concentrated to a quarter of its volume and chromatographed using Dowex® 50X8-200 resin to give the free amino acid (3R,4R)-4 (38 mg, 68% from 53) as a pale yellow solid, with spectroscopic data identical to that reported above.

Preparation of (3S,4R)-4-aminopyrrolidine-3-carboxylic acid hydrochloride 7.HCl

Following general procedure 4, Pd(OH)₂ on C (20 mg) was added to a stirred degassed solution of $(3S, 4R, \alpha S)$ -54 (100 mg, 0.26 mmol) in a mixture of EtOH (2 ml), H₂O (1.0 ml) and HCl (conc, 0.5 ml). The reaction mixture was stirred under H₂ (5 atm) overnight, then filtered through Celite® and concentrated *in vacuo* to give the crude (3S, 4R)- β -amino ester as a yellow oil (37 mg). This material was subsequently re-dissolved in HCl (1 M aq, 5 ml) and heated at 70 °C for 4 h. Following cooling to room temperature, the solution was washed with Et₂O and the aqueous phase concentrated in vacuo to afford the trans amino acid hydrochloride salt (3S,4R)-7.HCl (24 mg, 71% from 54) as a clear glass; $[a]_{D^{25}}$ +22.9 (c 0.50, H₂O); v_{max} (film) 3592–2333br s, 3405s, 2924s, 1729s, 1606m, 1544m, 1515s, 1455m, 1394s, 1296m, 1260m, 1221s, 1197s, 1130m, 843m; $\delta_{\rm H}$ (400 MHz, D₂O) 3.37-3.47 (2H, m, C(3)H and C(5)H_A), 3.58 (1H, dd, J 12.0, 7.2, C(2)H_A), 3.69 (1H, dd, J 12.0, 8.4, C(2)H_B), 3.79 (1H, dd, J 12.4, 8.8, C(5) $H_{\rm B}$), 4.25 (1H, q, J 6.8, C(4)H); $\delta_{\rm C}$ (100 MHz, CDCl₃) 40.6 (C(3)), 46.6 (C(5)), 47.9 (C(2)), 51.1 (C(4)), 176.1 (*C*O₂H); *m*/*z* (ESI⁺) 131 (MH⁺, 100%); HRMS, found 131.0821; $C_5H_{11}N_2O_2$ (MH⁺) requires 131.0821.

Acknowledgements

The authors wish to thank Pfizer for an industrial CASE awards (J. M. W) and New College, Oxford for a Junior Research Fellowship (A. D. S).

References and notes

- 1 M. Konishi, M. Nishio, K. Saitoh, T. Miyaki, T. Oki and H. Kawaguchi, J. Antibiot., 1989, 42, 1749; T. Oki, M. Hirano, K. Tomatsu, K. Numata and H. Kamei, J. Antibiot., 1989, 42, 1756; H. Ohki, Y. Inamoto, K. Kawabata, T. Kamimura and K. Sakane, J. Antibiot., 1991, **44**, 546.
- 2 J. Mittendorf, F. Kunisch, M. Matzke, H.-C. Militzer, A. Schmidt and W. Schönfeld, Biorg. Med. Chem. Lett., 2003, 13, 433.
- 3 P. Thorbeck, H. Hjeds and K. Schaumburg, Acta Chem. Scand., 1981, **B 35**, 473.

- 4 G. T. Wang, Y. Chen, S. Wang, R. Gentles, T. Sowin, W. Kati, S. Muchmore, V. Giranda, K. Stewart, H. Sham, D. Kempf and W. G. Laver, J. Med. Chem., 2001, 4, 1192; V. Stoll, K. D. Stewart, S. Muchmore, V. Giranda, Y. Y. Gu, G. Wang, C. J. Maring, Y. Chen, M. Sun, C. Zhao, A. L. Kennedy, D. L. Madigan, Y. Xu, A. Saldivar, W. Kati, G. Laver, T. Swoin, H. L. Sham, J. Greer and D. Kempf, Biochemistry, 2003, 42, 718.
- 5 X. Wang, J. F. Espinosa and S. H. Gellman, J. Am. Chem. Soc., 2000, 122, 4821; for related studies on β -peptides containing sulfonylated pyrrolidine residues see: H.-S. Lee, F. A. Syud, X. Wang and . H. Gellman, J. Am. Chem. Soc., 2001, 123, 7721
- 6 W. Tang, S. Wu and X. Zhang, J. Am. Chem. Soc., 2003, 125, 9570.
- R. Hanselmann, J. Zhou, P. Ma and P. N. Confalone, J. Org. Chem., 2003, 68, 8739.
- 8 V. K. Aggarwal, S. Roseblade and R. Alexander, Org. Biomol. Chem., 2003, 1, 684.
- 9 O. Miyata, K. Muroya, T. Kobayashi, R. Yamanaka, S. Kajisa, J. Koide and T. Naito, Tetrahedron, 2002, 58, 4459.
- 10 H.-S. Lee, P. R. LePlae, E. A. Porter and S. H. Gellman, J. Org. Chem., 2001, 66, 3597.
- 11 S. G. Davies, O. Ichihara, I. Lenoir and I. A. S. Walters, J. Chem. Soc., Perkin Trans. 1., 1994, 1411; for the asymmetric synthesis of derivatives of cispentacin using kinetic resolution methodology see: S. Bailey, S. G. Davies, A. D. Smith and J. M. Withey, Chem. Commun., 2002, 2910; S. G. Davies, D. Díez, M. M. El Hammouni, N. M. Garrido, A. C. Garner, M. J. C. Long, R. M. Morrison, A. D. Smith, M. J. Sweet and J. M. Withey, Chem. Commun., 2003, 2410
- 12 M. E. Bunnage, A. M. Chippindale, S. G. Davies, R. M. Parkin, A. D. Smith and J. M. Withey, Org. Biomol. Chem., 2003, 1, 3698.
- 13 M. A. Gianturco, P. Friedel and A. S. Giammarino, Tetrahedron, 1964, 20, 1763.
- 14 B. Dupré and A. I. Meyers, J. Org. Chem., 1991, 56, 3197.
 15 S. G. Davies, S. W. Epstein, A. C. Garner, O. Ichihara and A. D. Smith, Tetrahedron: Asymmetry, 2002, 13, 1555.
- 16 For example, see: Y. S. Park and P. Beak, J. Org. Chem., 1997, 62, 1574; M. D. Curtis and P. Beak, J. Org. Chem., 1999, 64, 2996; S. H. Lim, M. D. Curtis and P. Beak, Org. Lett., 2001, 3, 711; T. A. Johnson, D. O. Jang, B. W. Slafer, M. D. Curtis and P. Beak, J. Am. Chem. Soc., 2002, 124, 11689; P. Beak and W. K. Lee, J. Org. Chem., 1993, 58, 1109.
- 17 J. F. Costello, S. G. Davies and O. Ichihara, Tetrahedron: Asymmetry, 1994, 5, 3919
- 18 The relative $(3R,4R,3'R,4'R,\alpha S)$ configuration within 38 was readily assigned by analogy with the selectivity observed in the related addition of lithium dibenzylamide.
- 19 The facile nature of the retro-conjugate addition reaction of substrate 39 upon purification on silica and the stability of the tetrahydrofuran β -amino esters 36 and 37 is presumably due to the increased acidity of protons α - to a lactone carbonyl as compared to α - to an ester carbonvl.
- 20 J. E. Baldwin, J. Chem. Soc., Chem. Commun., 1976, 734; J. E. Baldwin, J. Cutting, W. Dupont, L. Kruse, L. Silberman and R. C. Thomas, J. Chem. Soc., Chem. Commun., 1976, 736; J. E. Baldwin and L. I. Kruse, J. Chem. Soc., Chem. Commun., 1976, 233; J. E. Baldwin and M. J. Lusch, Tetrahedron, 1982, 38, 2939.
- 21 For example, see: P. J. De Clercq, Chem. Rev., 1997, 97, 1755.
- 22 As shown by ¹⁹F and ¹H NMR spectroscopic analysis of the derived methyl esters and subsequent derivatisation with both racemic and 99% e.e. Mosher's acid chloride.
- 23 A. Hosomi, Y. Sakata and H. Sakurai, Chem. Lett., 1984, 1117; Y. Terao, H. Kotaki, N. Imai and K. Achiwa, Chem. Pharm. Bull., 1985, 33, 2762; this compound is commercially available from Sigma-Aldrich.
- 24 M. Muto, T. Saito, Y. Tani, T. Akiba and K. Nakayama, Eur. Pat. Appl., 2003, EP 1310487.
- 25 M. A. Gianturco, P. Friedel and A. S. Giammarino, Tetrahedron, 1964, 20, 1763.
- 26 D. J. Watkin, C. K. Prout, J. R. Carruthers, P. W. Betteridge and R. I. Cooper, CRYSTALS, 2001, Issue 11, Chemical Crystallography Laboratory, Oxford, UK.
- 27 Commercially available from Sigma-Aldrich; see also F. Sondheimer, N. Stjernstrom and D. Rosenthal, J. Org. Chem., 1959, 24, 1280.